

Literature Report

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Copper-Catalyzed Direct Amination of Quinoline N-Oxides via C–H Bond Activation under Mild Conditions

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Optimizing Reaction Parameters for the Condensation of Quinoline N-Oxide 1a with Piperidine 2a



entry	catalyst (equiv)	solvent	amine (equiv)	yield ^b (9	6 entry	catalyst (equiv)	solvent	amine (equiv)	yield ^b (%)
1	$Cu(OAc)_{2}$ (0.2)	toluene	8.0	85	13	CuI (0.2)	DCE	8.0	NR
2	$Cu(OTf)_2$ (0.2)	toluene	8.0	84	14	CuI (0.2)	DMF	8.0	trace
3	$CuBr_2$ (0.2)	toluene	8.0	83	15	CuI (0.2)	toluene	7.0	85
4	CuBr (0.2)	toluene	8.0	80	16	CuI (0.2)	toluene	6.0	82
5	CuCl (0.2)	toluene	8.0	89	17	CuI (0.2)	toluene	4.0	75
6	CuI (0.2)	toluene	8.0	94	18	CuI (0.1)	toluene	8.0	91
7	$NiCl_2 \cdot 6H_2O(0.2)$	toluene	8.0	NR	19	CuI (0.08)	toluene	8.0	86
8	$Pd(OAc)_2$ (0.2)	toluene	8.0	NR	20	CuI (0.05)	toluene	8.0	78
9	$CoCl_2$ (0.2)	toluene	8.0	NR	21	CuI (0.1)	toluene	8.0	86 ^c
10	CuI (0.2)	THF	8.0	80	22	CuI (0.1)	toluene	8.0	70^d
11	CuI (0.2)	CH ₃ CN	8.0	78	^a Reaction	on conditions: 1a (0).2 mmol), so	lvent (1.5 mL),	50 °C, 7 h
12	CuI (0.2)	DMSO	8.0	NR	^b Isolate	d yield based on 1a	. ^c 40 °C. ^{´d} 30	°C. NR = no re	eaction.

Cul (10 mol %), amine (8.0 equiv), toluene, 50° C, under air



Copper-Catalyzed Amination of Quinoline N-Oxide with Various Amines





Copper-Catalyzed Amination of Quinoline N-Oxides with Piperidine





Plausible Reaction Mechanism



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Copper-Catalyzed Direct Amination of Quinoline N-Oxides via C–H Bond Activation

Advantages:

 – simple system, high efficiency, atomic economy, environmental friendliness

 low reaction temperature, and ligand, additives, base, and external oxidant free conditions

the feasibility of the mechanism was also verified by density functional theory (DFT)

Disadvantages: Mechanism, substrates, application, selectivity



Thank you!