



北京大学

PEKING UNIVERSITY

化学生物学与生物技术学院

School of Chemical Biology & Biotechnology

Decarboxylative Arylation of α -Amino Acids via Photoredox Catalysis

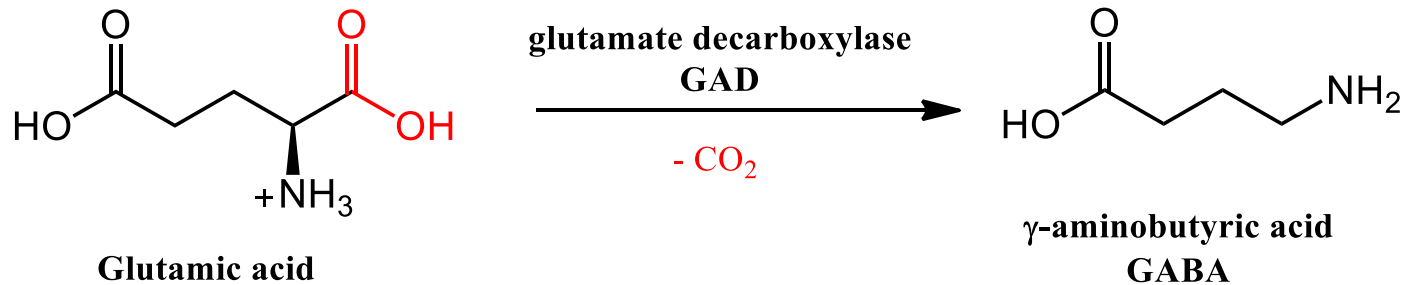
Reporter: Lu YanBo

Supervisor: Prof. Zhao Jing

Dr. Hong Mei

Decarboxylation

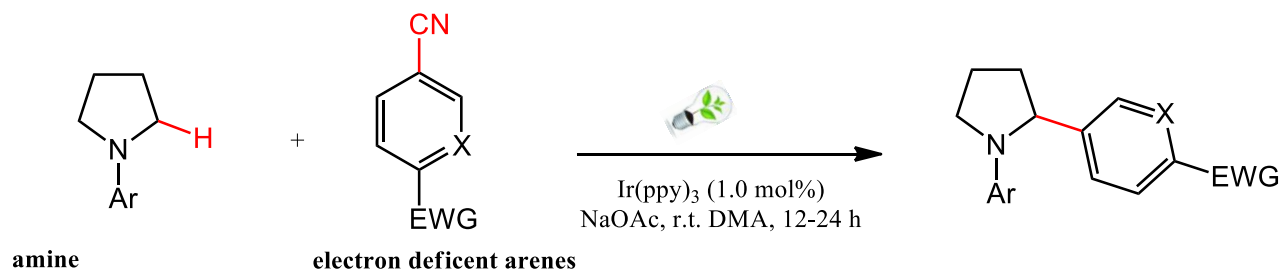
- Biochemical Precedent



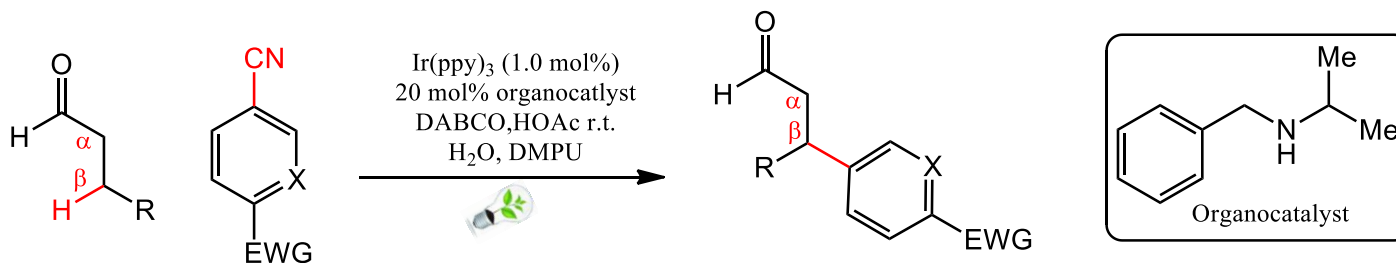
- Chemical Precedents

- Hunsdiecker, Kolbe, Barton decarboxylation
- Coupling reaction (Myers and Gooßen)

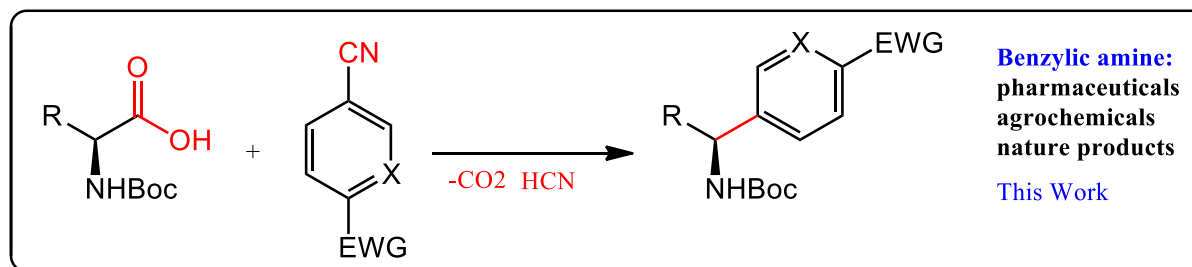
Previous Work



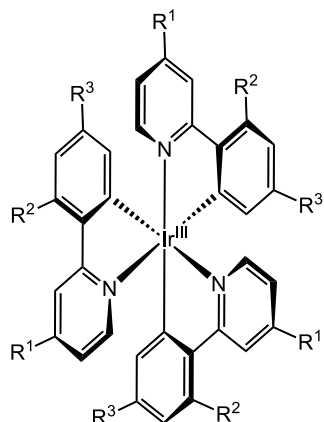
McNally, A.; Prier, C. K.; MacMillan, D. W. C. *Science* **2011**, 334, 1114.



Pirnot, M. T.; Rankic, D. A.; Martin, D. B.; MacMillan, D. W. *Science* **2013**, 339, 1593.



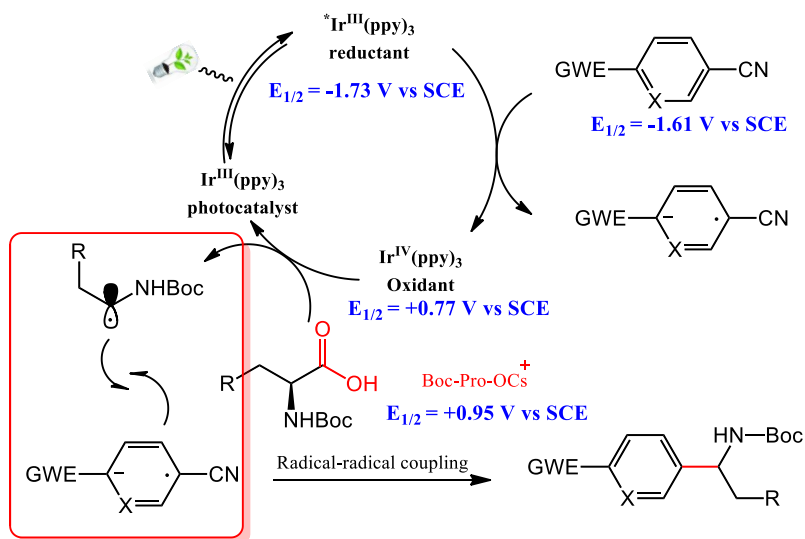
Ligand effects on ground-state redox properties of catalysts



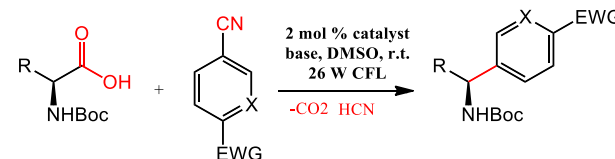
Group	Photocatalyst	$E_{1/2}^{\text{red}}[\text{Ir}^{\text{IV}}/\text{Ir}^{\text{III}}]$ vs SCE (Oxidant)	$E_{1/2}^{\text{red}}[\text{Ir}^{\text{IV}}/\text{Ir}^{\text{III}}]$ vs SCE (Reductant)	Entry
$R^1 = R^2 = R^3 = \text{H}$	$\text{Ir}(\text{ppy})_3$	+ 0.77 V	- 1.73 V	A
$R^1 = R^2 = \text{H}, R^3 = \text{F}$	$\text{Ir}(p\text{-F-ppy})_3$	+ 0.97 V	- 1.60 V	B
$R^1 = \text{H}, R^2 = R^3 = \text{F}$	$\text{Ir}(\text{dFppy})_3$	+ 1.13 V	- 1.44 V	C
$R^1 = t\text{-Bu}, R^2 = \text{H}, R^3 = \text{F}$	$\text{Ir}[p\text{-F}(t\text{-Bu-ppy})]_3$	---	---	D
$R^1 = t\text{-Bu}, R^2 = R^3 = \text{F}$	$\text{Ir}[\text{dF}(t\text{-Bu-ppy})]_3$	---	- 1.67 V	E
	$\text{Ir}[\text{dF}(\text{CF}_3)\text{ppy}]_3(\text{dtbpy})^+$	+ 1.69 V	- 0.89 V	F

Tucker, J. W.; Stephenson, C. R. J. *J. Org. Chem.* **2012**, *77*, 1617.

Design

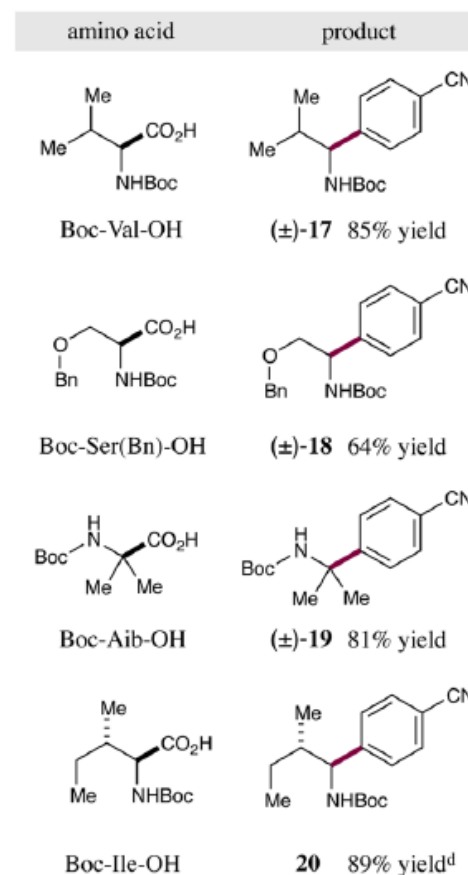
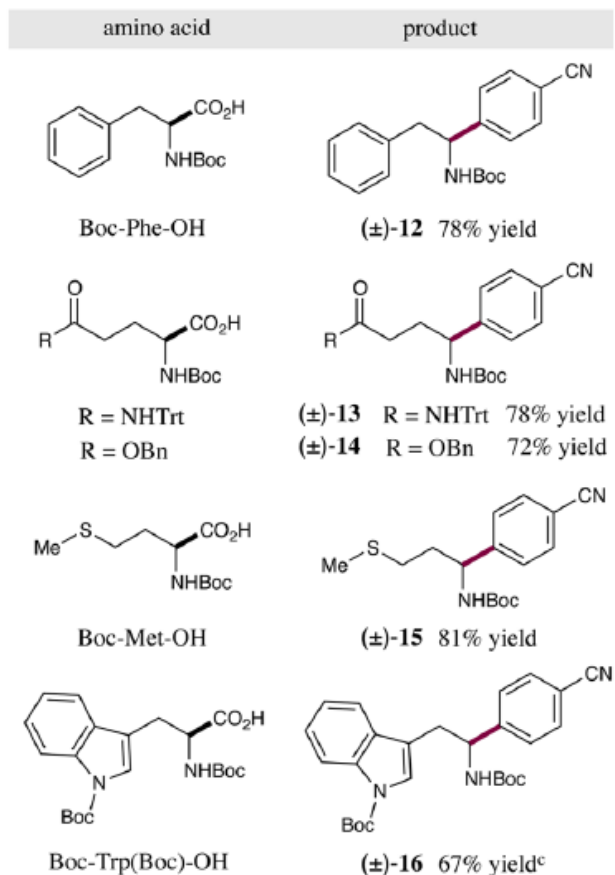
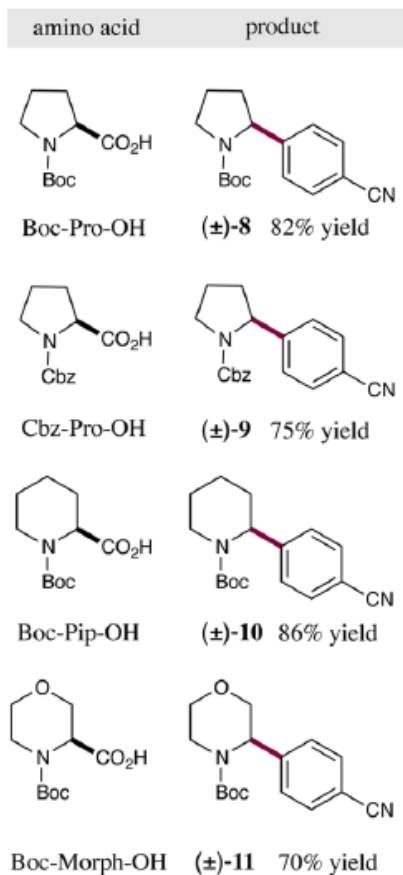


Conditions Screening

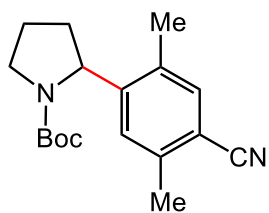


entry	photocatalyst	base	% yield ^a
1	$\text{Ir}(\text{ppy})_3$	K_2HPO_4	12
2	$\text{Ir}(p\text{-F-ppy})_3$	K_2HPO_4	58
3	$\text{Ir}(\text{dFppy})_3$	K_2HPO_4	54
4	$\text{Ir}[\text{dF}(\text{CF}_3)\text{ppy}]_2(\text{dtbpy})$	K_2HPO_4	trace
5	$\text{Ir}[p\text{-F}(t\text{-Bu-ppy})]_3$	K_2HPO_4	73
6	$\text{Ir}[\text{dF}(t\text{-Bu-ppy})]_3$	K_2HPO_4	68
7	$\text{Ir}[p\text{-F}(t\text{-Bu-ppy})]_3$	CsF	83
8 ^b	$\text{Ir}[p\text{-F}(t\text{-Bu-ppy})]_3$	CsF	0
9	none	CsF	0
10	$\text{Ir}[p\text{-F}(t\text{-Bu-ppy})]_3$	none	trace

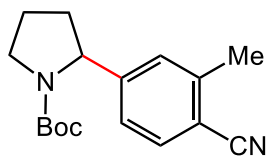
α -Amino Acid Scope



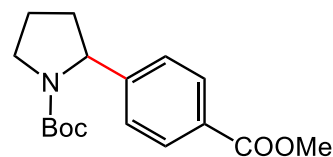
Aromatic Scope



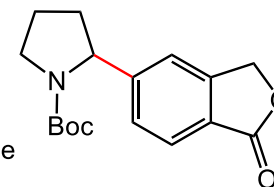
2a 85% yield



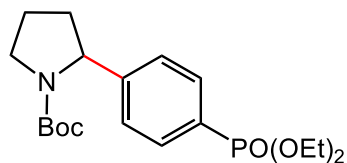
2b 77% yield



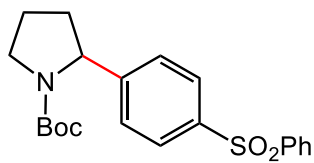
2c 52% yield



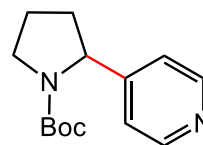
2d 70% yield



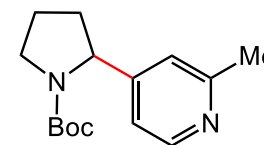
2e 65% yield



2f 64% yield

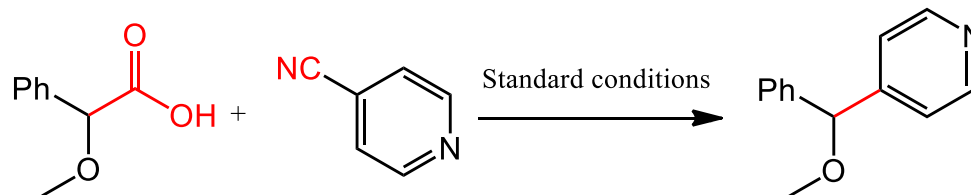
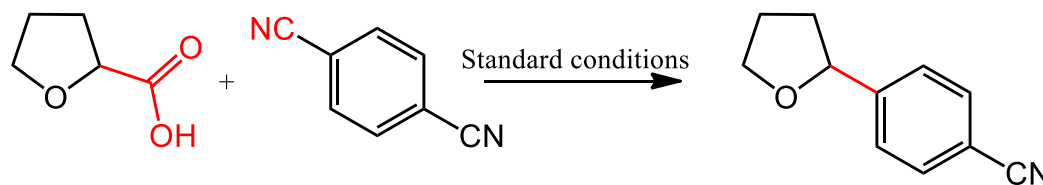


2g 73% yield



2h 83% yield

Another Two Examples



Summary

- Ligand effects are important for Photocatalyzed reactions.
- The substrates are limited in electron-deficient aromatic compounds.