

Review Presentation about Mercury Biosensor



Reporter: zouyan
Advisor: Prof Zhao
Dr. Wei



Background

- ❖ **Mercury:** a liquid metal with a high thermal conductivity, has many unique properties that make it useful in more than 3000 industrial applications, ranging from **fungicides and bactericides** in the agriculture industry(alkyl mercury compounds) to **heat transfer agents** and **catalysts** in the chemical industry.





Background

- ❖ However, mercury, as a highly toxic compound, caused heavy environmental contamination and deleterious effects on human health.



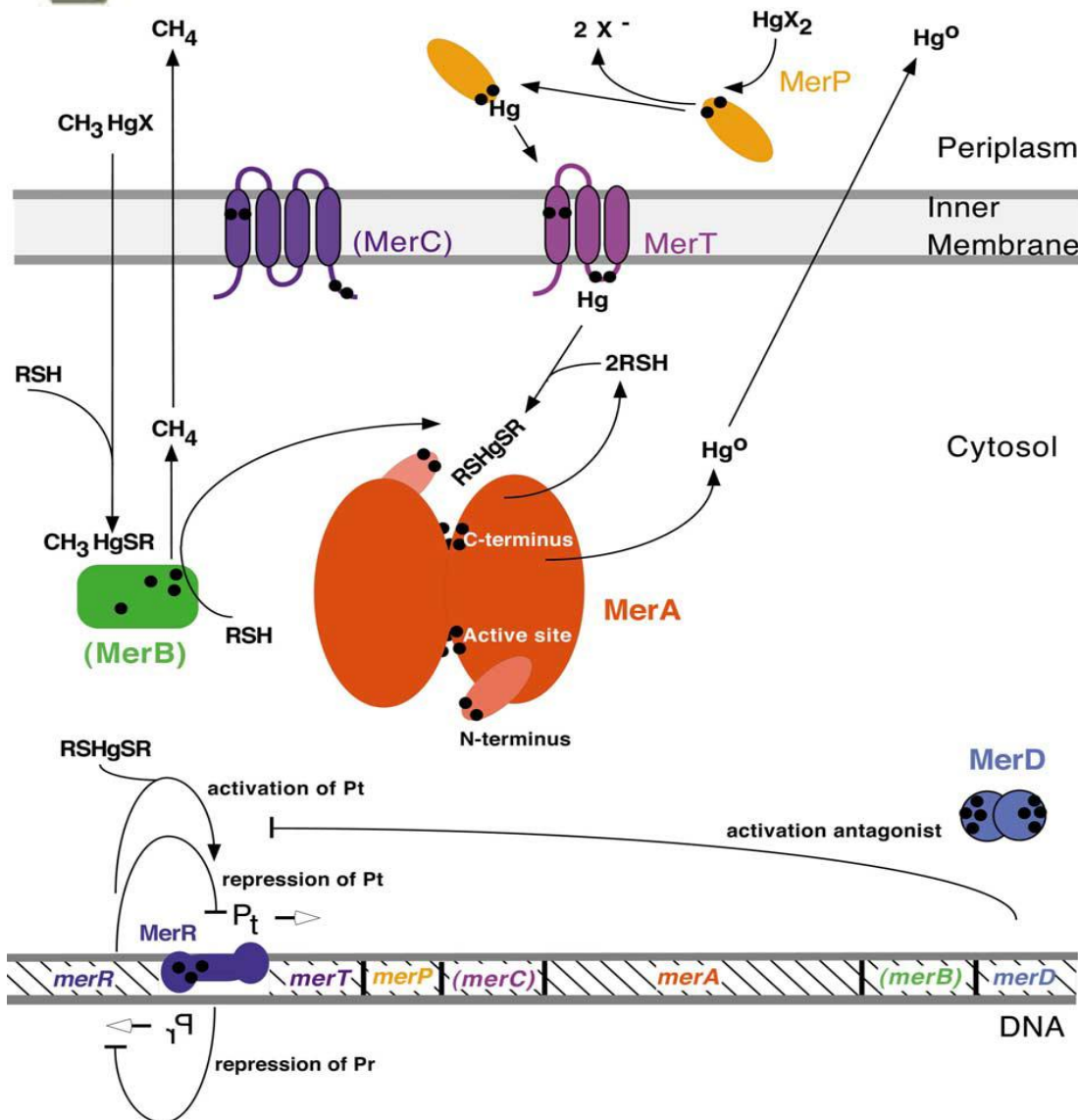


Detection method of mercury

- ❖ **Physical:** AAS, AES, GC-MS
- ❖ **Chemical:** inductively coupled plasma spectrometry
electrochemical method
disadvantage: expensive instruments
complicated operation
secondary pollution
- ❖ **Biosensor:** whole-cell sensor
protein-based sensor
oligonucleotide-Based



Mercury Resistance System



narrow-spectrum mercury resistance (lack B) :
inorganic mercury

broad-spectrum mercury resistance :
inorganic mercury &
organic mercury

merR
↓
merTP(C)A(B)D

Anal. Chem. **1995**, *67*, 667–669

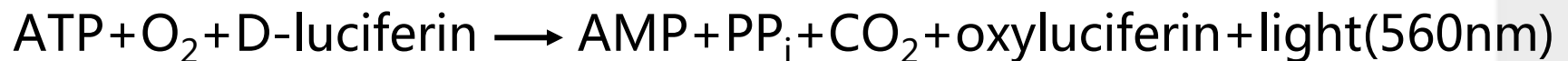
A Luminescence-Based Mercury Biosensor

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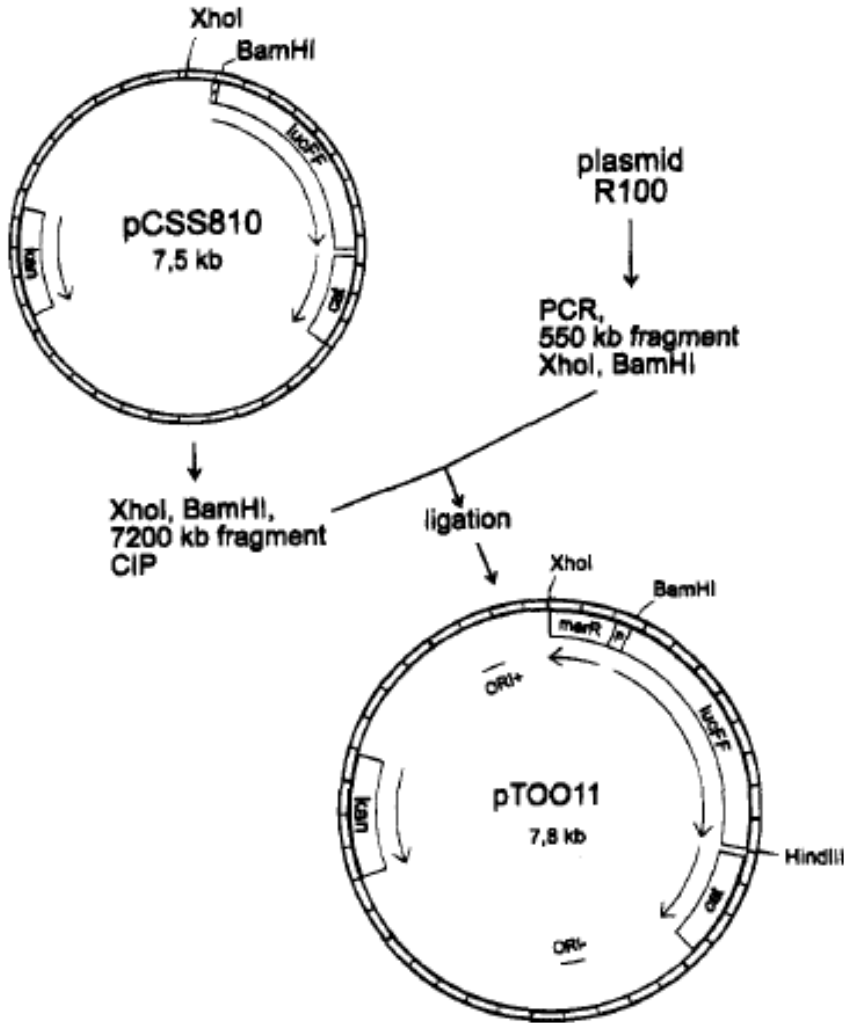
Biosensor: mer operon — firefly luciferase

The firefly luciferase catalyses the reaction:





Mechanism



PmerR :

F: 5'-TTAAGGATCCCCTCATAGTTAATTTCTCCTC
TTTTGGATTGGATAG-3'

R: 5'-ATATCTCGAGCTAAGGCATAGCTGACCT-3'

electroporation

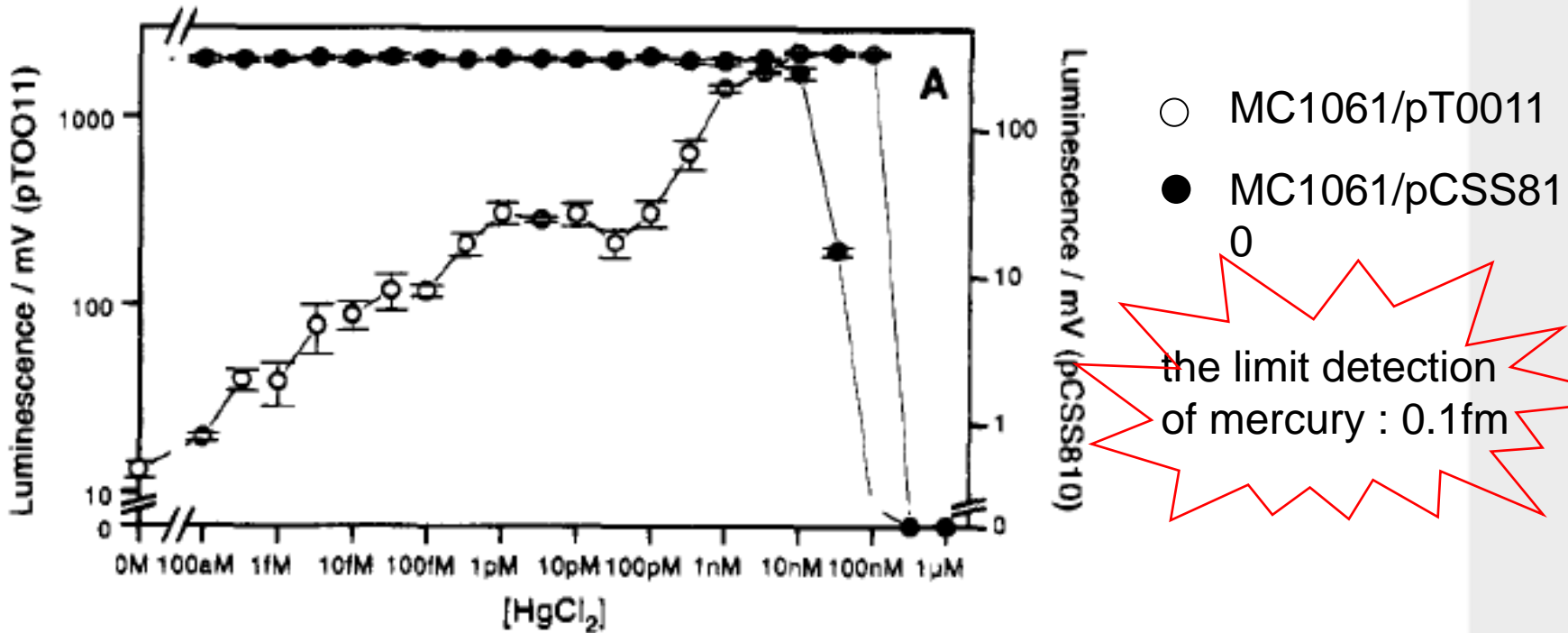


E.coli MC1061 cell





Results

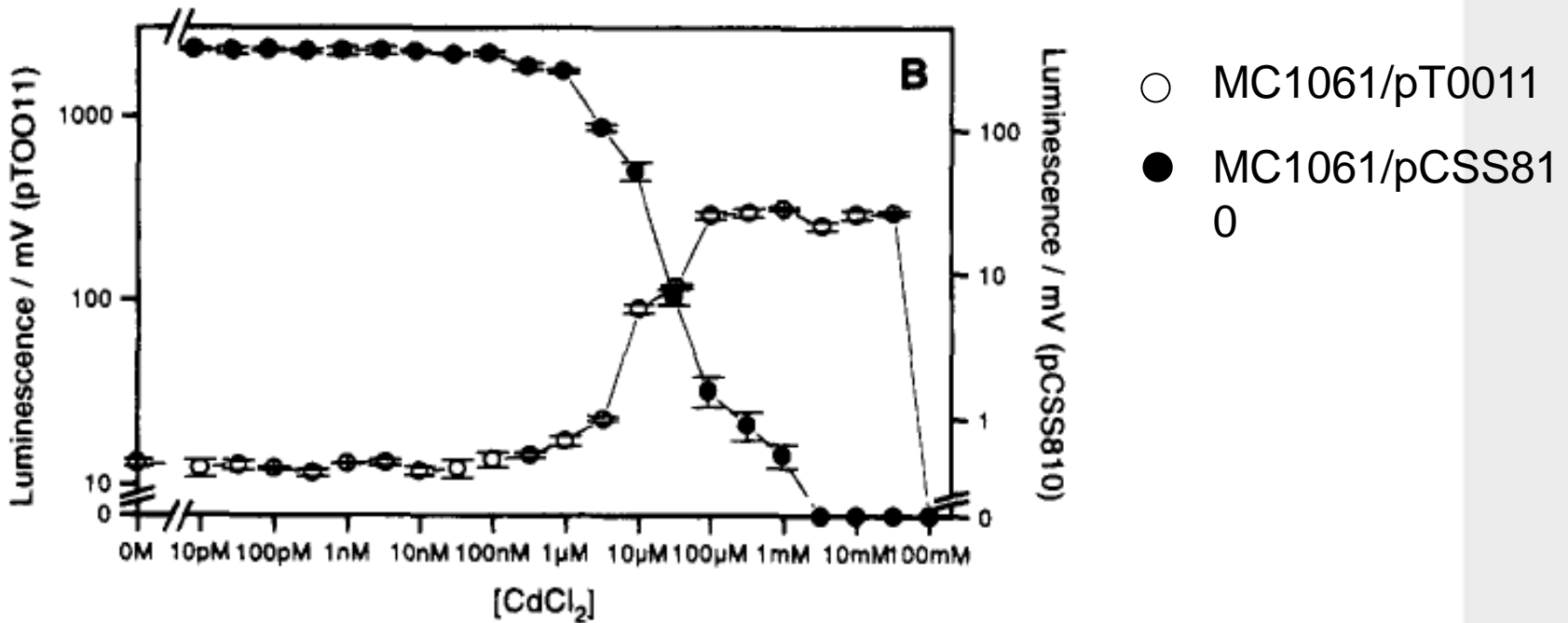


- ❖ The luminescence increased with increasing HgCl_2 concentration in a linear manner to concentration of $0.1\mu\text{M}$.





- ❖ the only ion that was observed to interfere with the measurement was cadmium



- ❖ Cd induced remarkable luminescence too, however, the concentration needed for induction was almost 10^7 -fold that of Hg.





Anal. Chem. **2001**, *73*, 5168–5171

Detection of Organomercurials with Sensor Bacteria

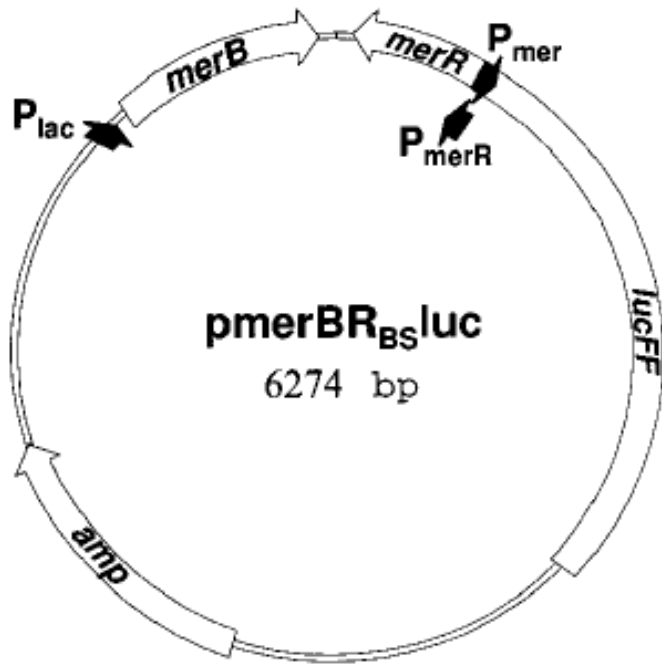
Angela Ivask,^{†,‡} Kaisa Hakkila,[‡] and Marko Virta^{*,‡}

National Institute of Chemical Physics and Biophysics, Tallinn, Estonia, and Department of Biotechnology, University of Turku, Turku, Finland





Mechanism



receptor — reporter (lucFF)

PmerB:

5'-ATATCACGTGATGAAGCTCGCCCCATATAT-3'

5'-AATTCTCGAGGCTGCGAATCCGATGCCGGT-3'

PmerR(mer):

5'-ATATCTCGAGTCCTCAGCATAGTACCGGGA-3'

5'-TTAAGGATCCCCTCATACGCTTGTCCTTTCAA-3'

Structure of the plasmid pmerBRBSluc

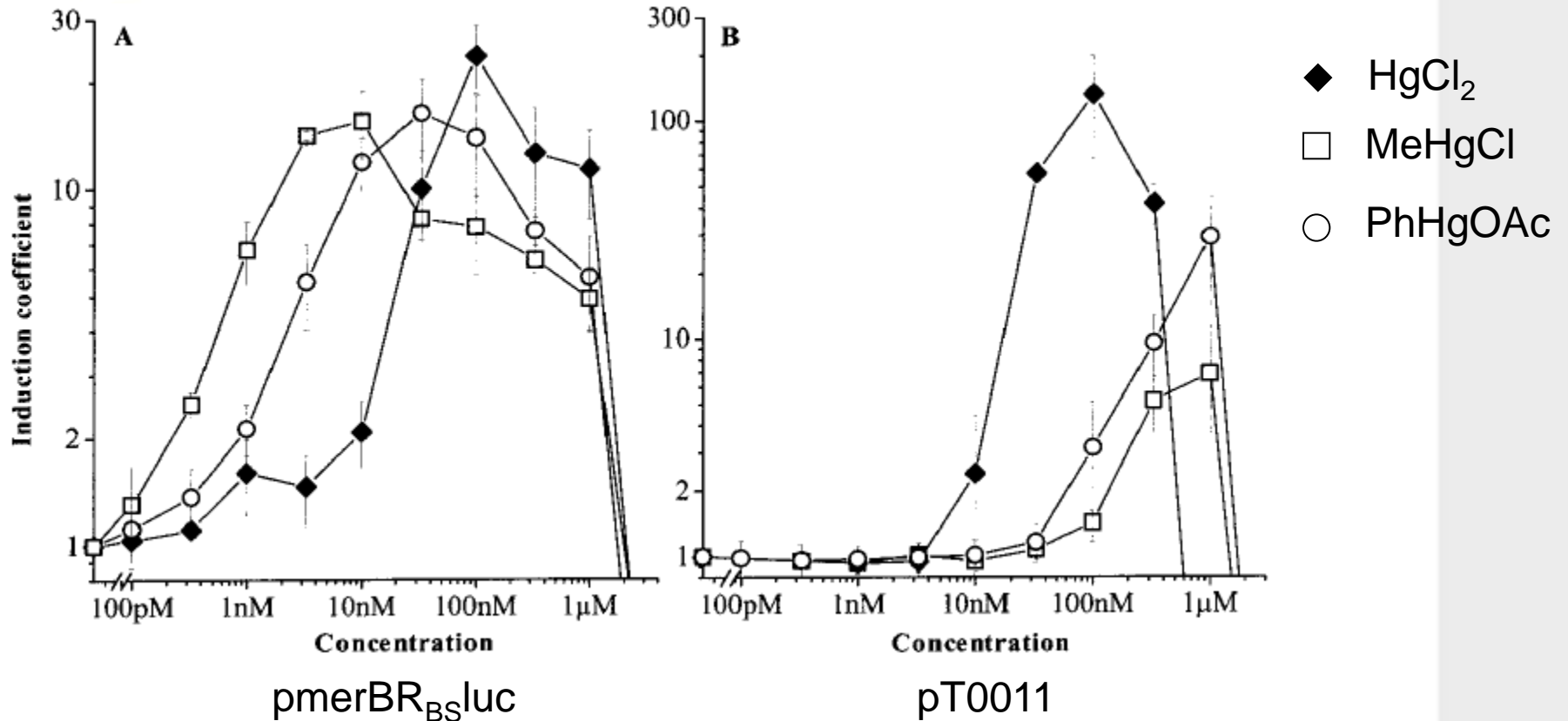


Escherichiacoli MC1061



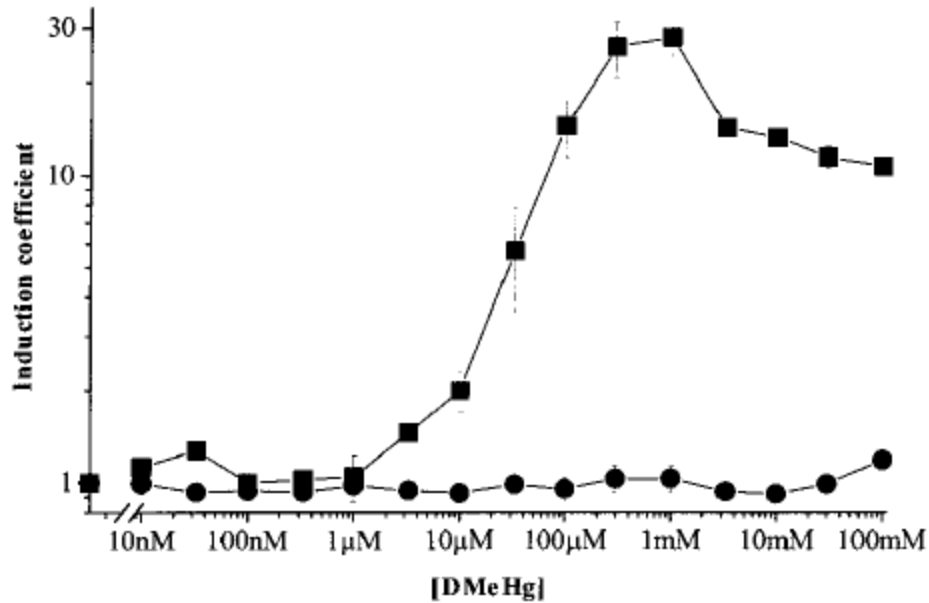


Results



A: The lowest concentration that caused noticeable response was HgCl₂ (10 nM), MeHgCl (0.2 nM) and PhHgOAc (1 nM) .

B: The concentrations were 10nM (HgCl₂), 0.1nM (MeHgCl and PhHgOAc).



- MC1061(pmerBRBSluc)
- MC1061(pTOO11)

❖ Response to DMeHg, but the increase of luminescence was seen only with MC1061(pmerBRBSluc).





Protein-Based Hg(II) Detectors

J|A|C|S
COMMUNICATIONS

Published on Web 03/03/2007

Design of an Emission Ratiometric Biosensor from MerR Family Proteins: A Sensitive and Selective Sensor for Hg²⁺

Seraphine V. Wegner, Ayse Okesli, Peng Chen, and Chuan He*

Department of Chemistry, The University of Chicago, 5735 South Ellis Avenue, Chicago, Illinois 60637

Received November 21, 2006; Revised Manuscript Received February 15, 2007; E-mail: chuanhe@uchicago.edu

(19) **United States**

(12) **Patent Application Publication**
Lu et al.

(10) **Pub. No.: US 2011/0123982 A1**
(43) **Pub. Date: May 26, 2011**

(54) **NUCLEIC ACID BASED FLUORESCENT
SENSOR FOR COPPER DETECTION**

(76) Inventors: **Yi Lu**, Champaign, IL (US);
Juewen Liu, Kitchener (CA)

(21) Appl. No.: **12/598,070**

(22) PCT Filed: **Jul. 16, 2008**

(86) PCT No.: **PCT/US2008/070177**

§ 371 (c)(1),
(2), (4) Date: **Dec. 22, 2010**

Related U.S. Application Data

(60) Provisional application No. 60/950,062, filed on Jul. 16, 2007.

Publication Classification

(51) **Int. Cl.**
C12Q 1/68 (2006.01)
C07H 21/00 (2006.01)

(52) **U.S. Cl.** **435/6; 536/22.1**

(57) **ABSTRACT**

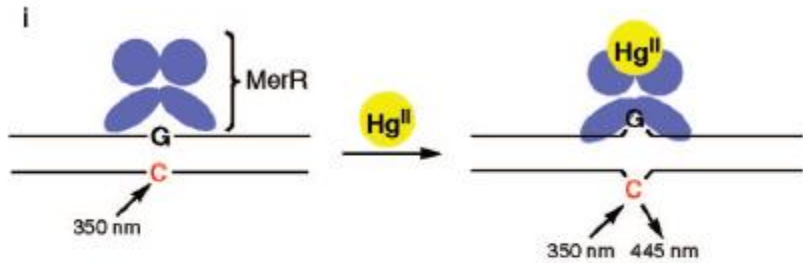
A nucleic acid enzyme responsive to copper, comprising an oligonucleotide comprising a nucleotide sequence of SEQ ID NO:1, wherein the nucleic acid enzyme is not self-cleaving.



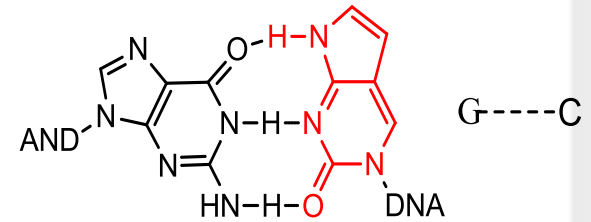


Mechanism

a) Turn-On



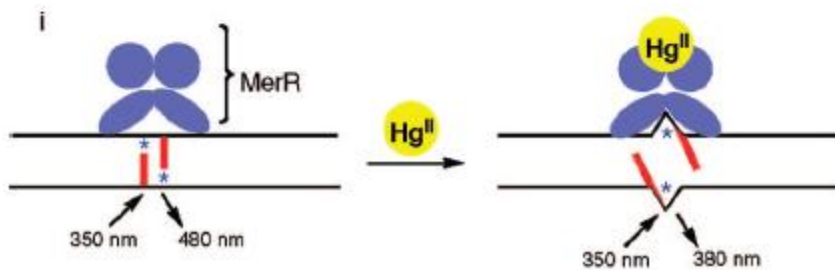
II:



III:

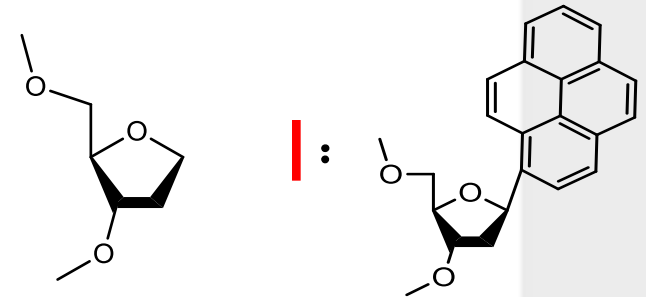
5'-CCTTGACTCCGTACATGAGTACGGAAGTCCC-3'
 3'-GGAAGT GAGGCATGTACTCATGCCTTCAGGG-5'

b) Ratiometric



II:

*: Ab:

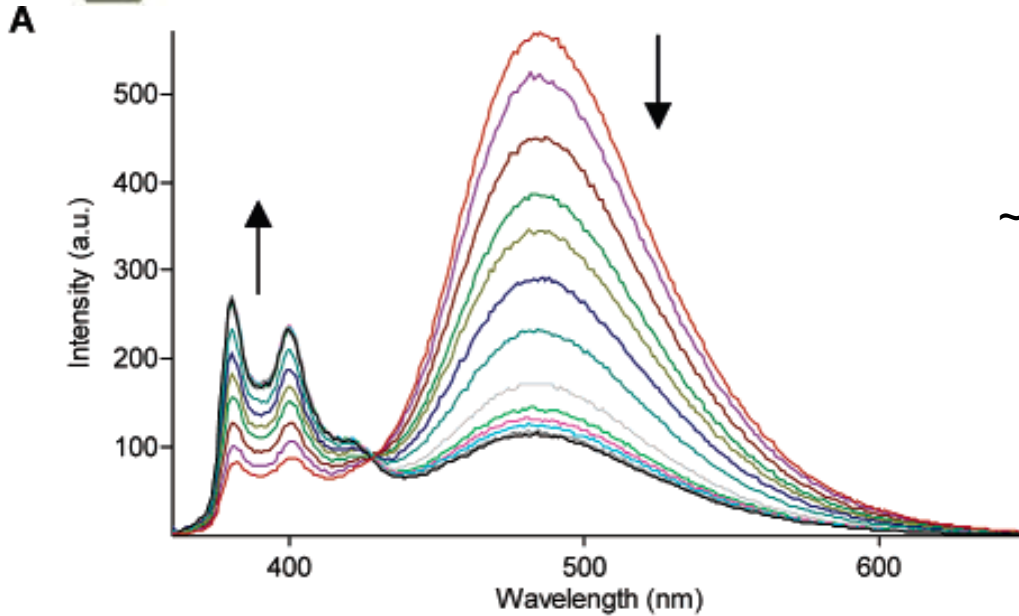


III:

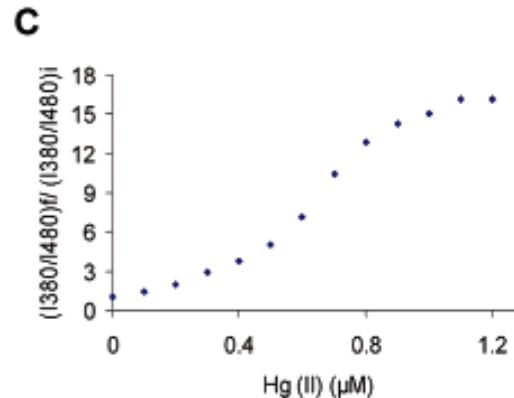
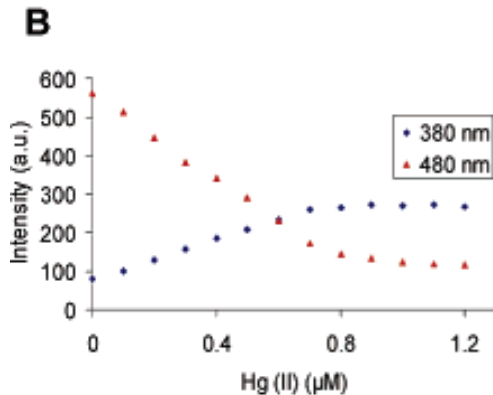
5'-CCTTGACTCCGTACA * AGTACGGAAGTCCC-3'
 3'-GGAAGT GAGGCATGT * TCATGCCTTCAGGG-5'



Results



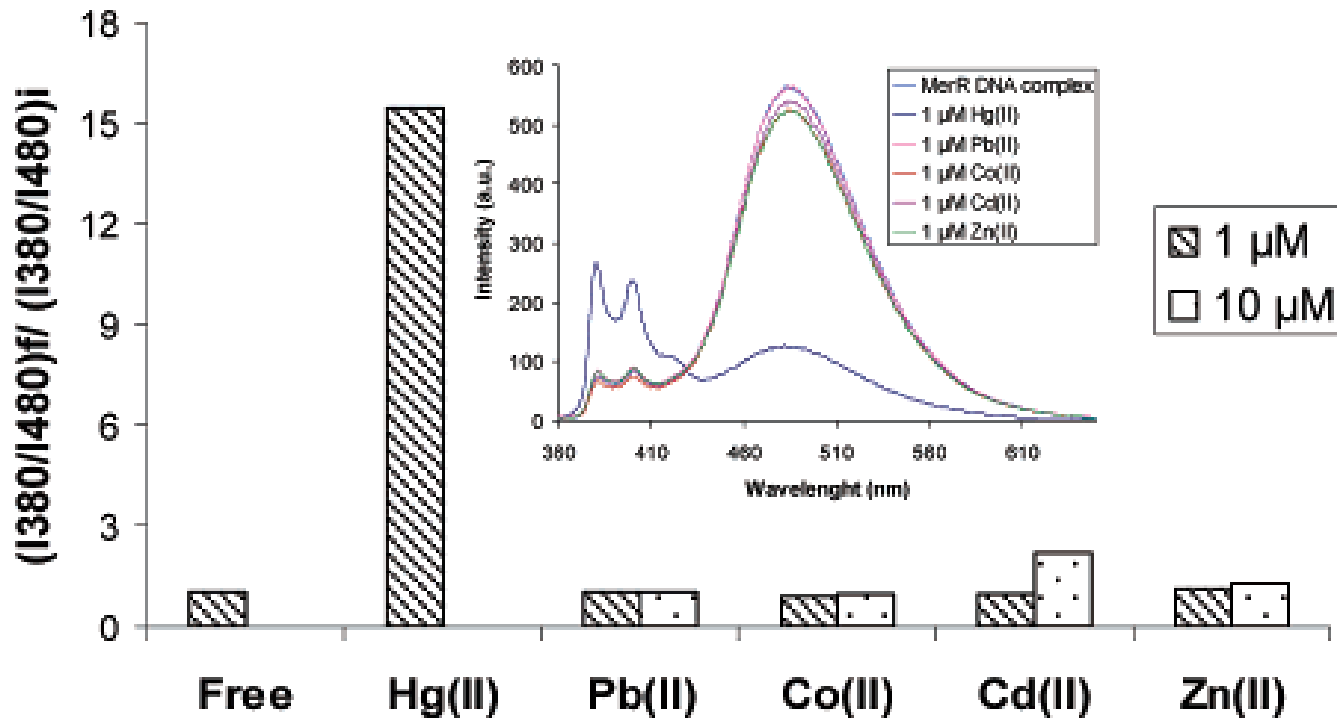
~5-fold decrease in pyrene eximer
~3.3-fold increase in pyrene monomer
↓
~15-fold ratiometric response



Fluorescence response of MerR-DNA complex to the addition of Hg^{2+} at 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, 0.8, 0.9, 1.0, 1.1, and 1.2 μM .



Results



Ratiometric response for different metal ions at 1 and 10 μM .





Oligonucleotide-Based Hg(II) Sensor

Molecular Sensors

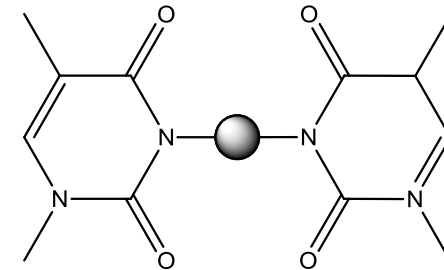
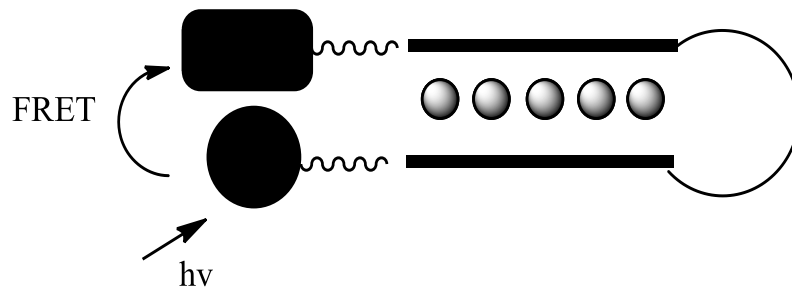
**Highly Selective Oligonucleotide-Based
Sensor for Mercury(II) in Aqueous
Solutions****



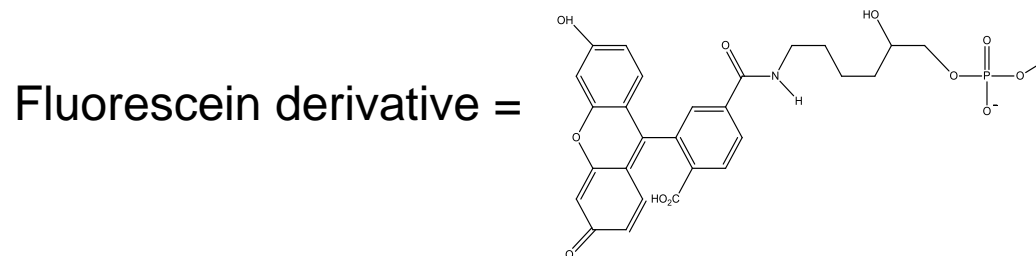
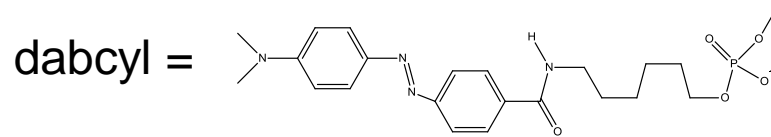


Mechanism

D-ODN-F: 5'-dabcyl-d($\overbrace{\text{TTCTTTCTTCCCCTT}}^{\text{Hg-binding sequence}}\overbrace{\text{TGTTTGTT}}^{\text{linker}}$)-($\overbrace{\text{6-fluorescein}}$)-3'

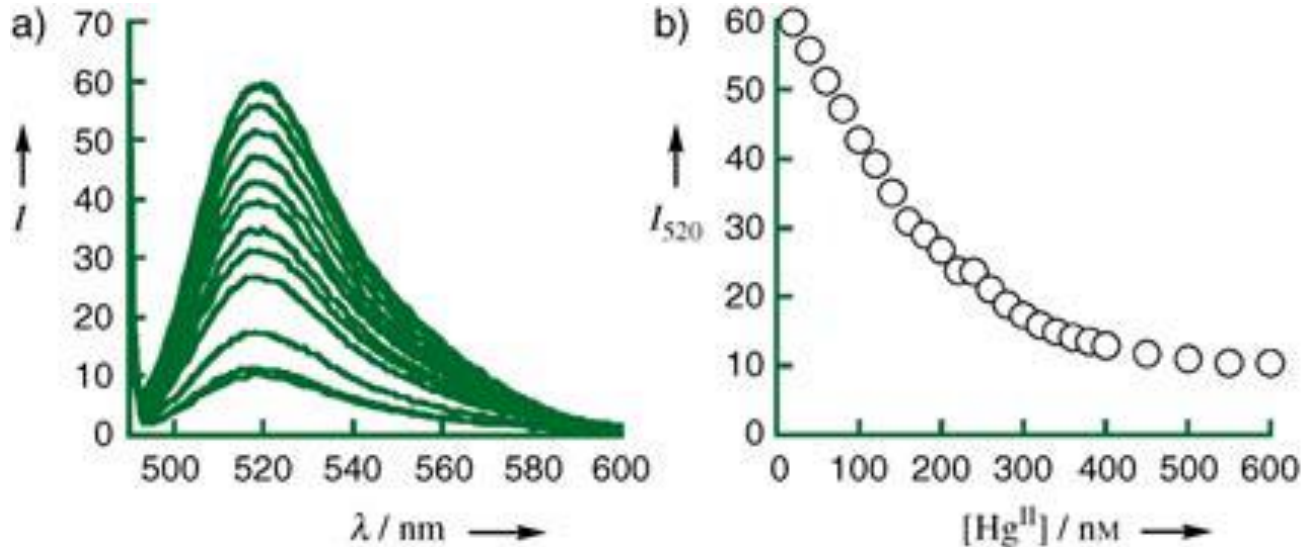


T-Hg-T





Results

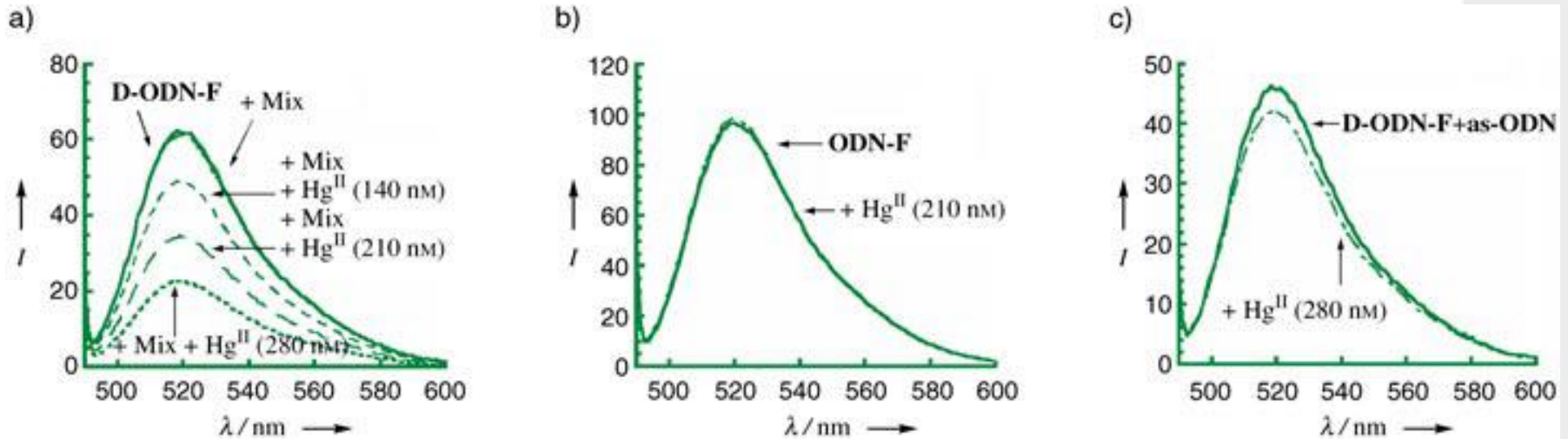


- ❖ A: Fluorescence response of D-ODN-F (10 nM) upon addition of Hg^{II} ion. The intensity of the fluorescence decreased as the concentration of Hg^{II} increased.
- ❖ B: Fluorescence emission intensity (520 nm) versus Hg^{II} concentration.





Results



- ❖ a: a mixture of heavy metal ions and appropriate concentration of $\text{Hg}(\text{ClO}_4)_2$.
- ❖ b: contrast of ODN-F
- ❖ A contrast of as-ODN(no D and F)





Thank You !

