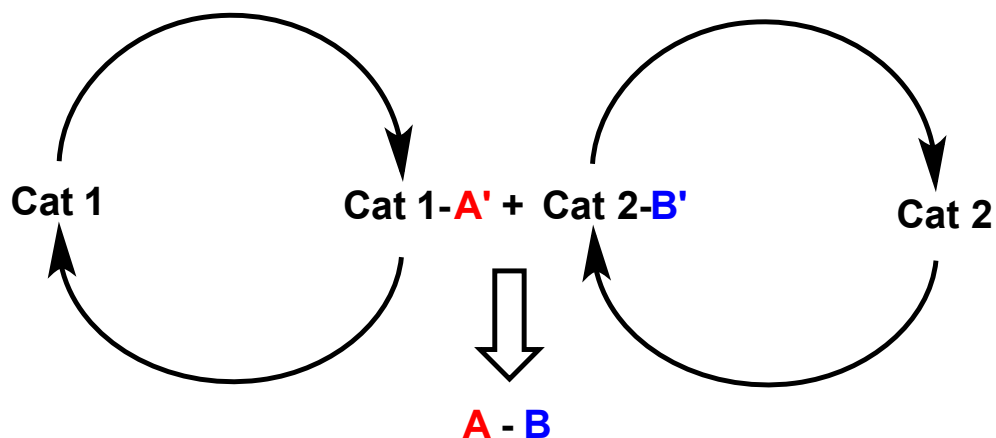


Cooperative Catalysis and Activation with N-Heterocyclic Carbenes



Michael H. Wang and Karl A. Scheidt*



*enhanced reactivity
greater yield*

*improved stereoselectivity
new chemo- and regioselectivity*

Reporter: Yang Li

Supervisor: Prof. Yong Huang

Mar. 27th, 2017

Outline:

1. Introduction

2. NHC Cooperative Catalysis

2.1 NHC/Lewis Acid Cooperative Catalysis

2.2 Oxidative NHC Catalysis

2.3 Oxidative NHC/Lewis Acid Cooperative Catalysis

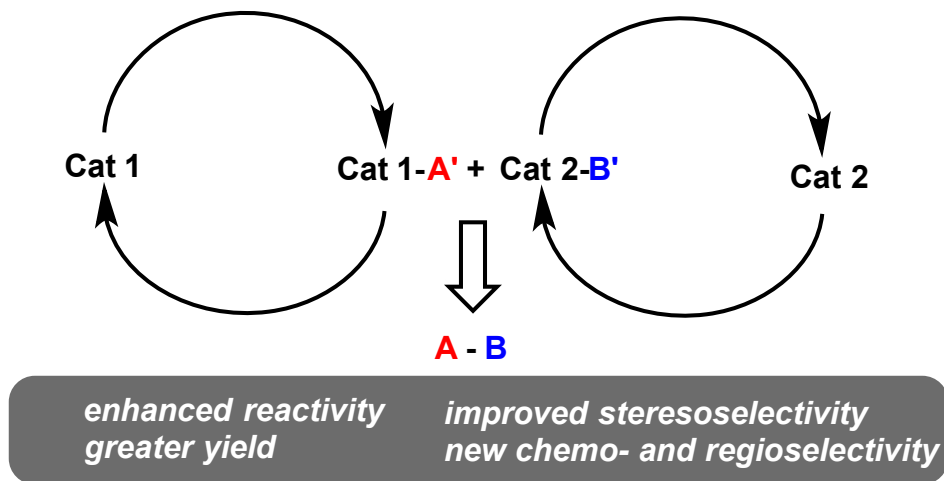
2.4 NHC/Brønsted Acid Cooperative Catalysis

2.5 NHC/Brønsted Base Cooperative Catalysis

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1.Introduction:



In cooperative (synergistic) systems, two distinct catalysts independently activate separate substrates to generate products with remarkable efficiency and selectivity that are unobtainable through single-catalyst systems

In many of the following examples, rigorous kinetic studies are not readily available to examine the impact of cocatalyst additives on reaction rate; however, addition of this second species led to observable improvements (e.g., yield, stereoselectivity), which arise owing to the enhanced rates of the favored pathways.

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2.4 NHC/Brønsted Acid Cooperative Catalysis

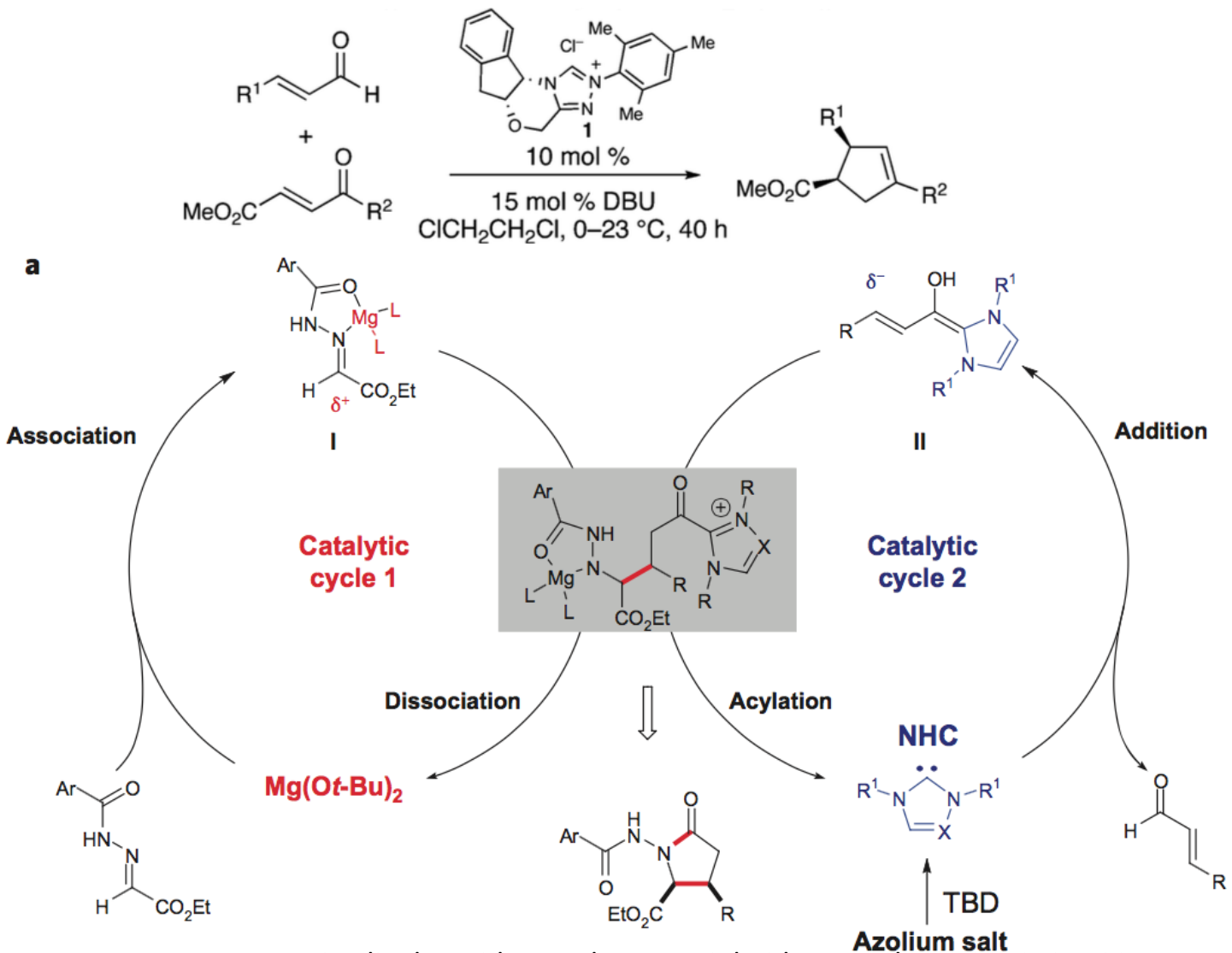
2.5 NHC/Brønsted Base Cooperative Catalysis

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2.1 NHC/Lewis Acid Cooperative Catalysis

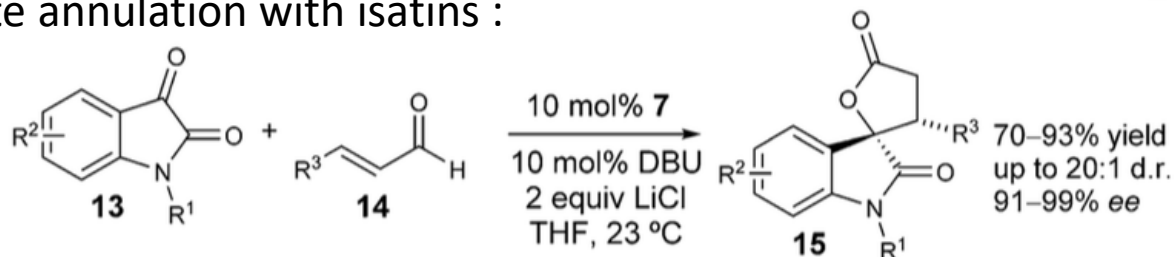
Enhanced reactivity by cooperative catalysis :



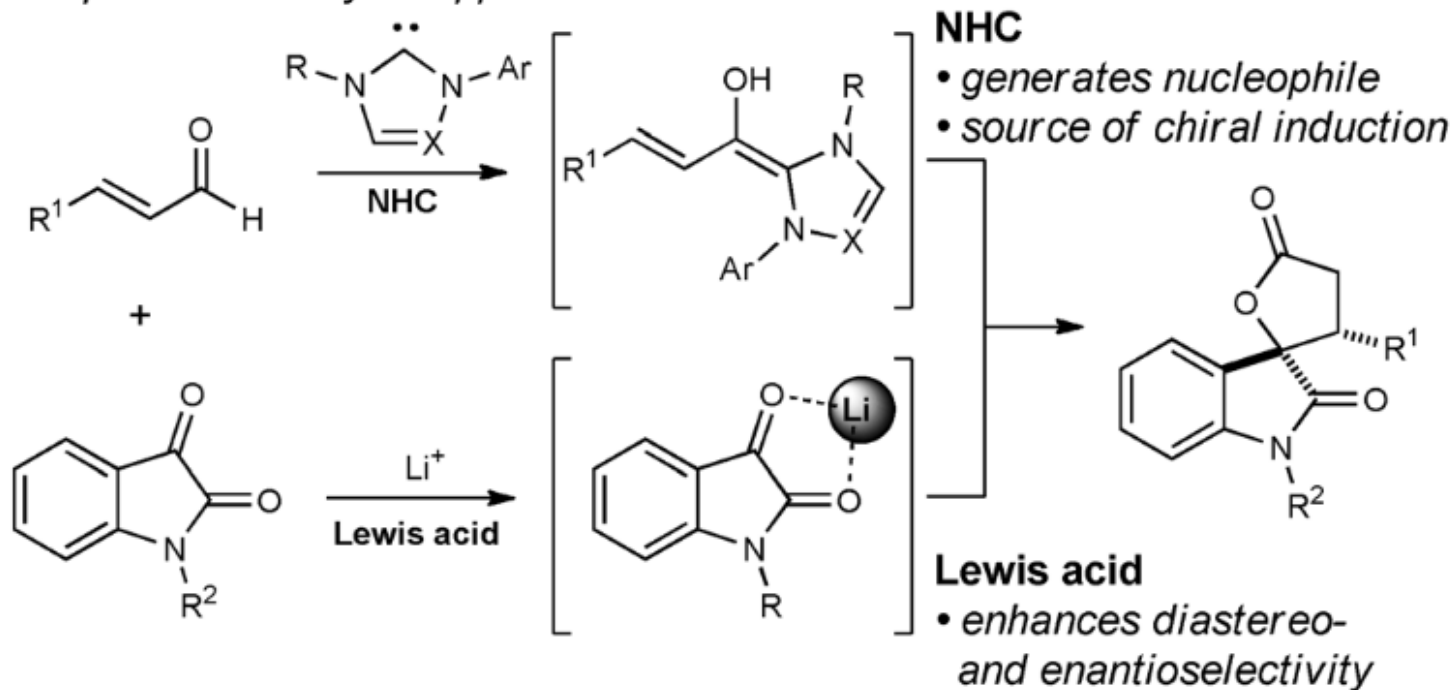
D. E. A. Raup, B. Cardinal-David, D. Holte, K. A. Scheidt, Nat. Chem. 2010, 2, 766.

2.1 NHC/Lewis Acid Cooperative Catalysis

Homoenolate annulation with isatins :

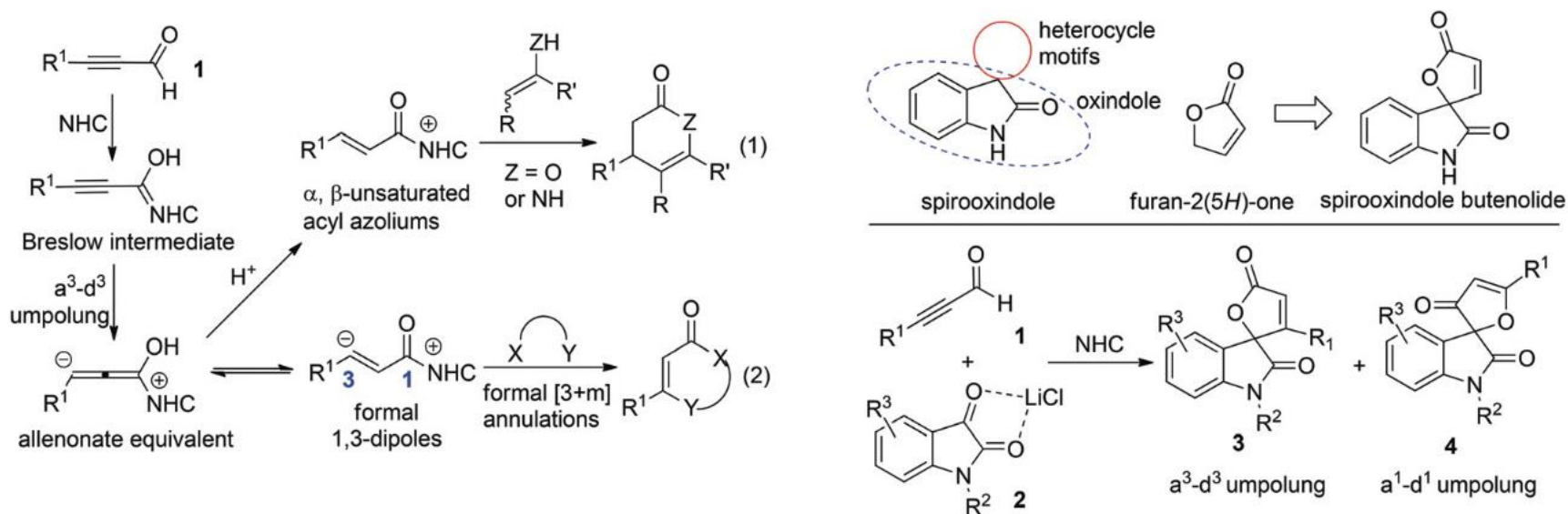
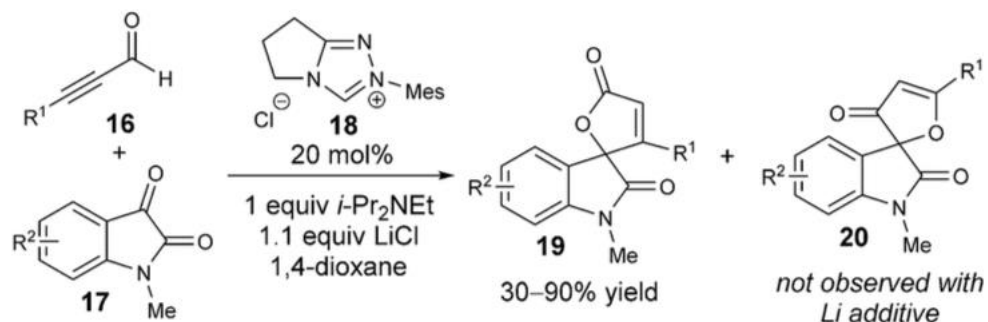


Cooperative catalysis approach

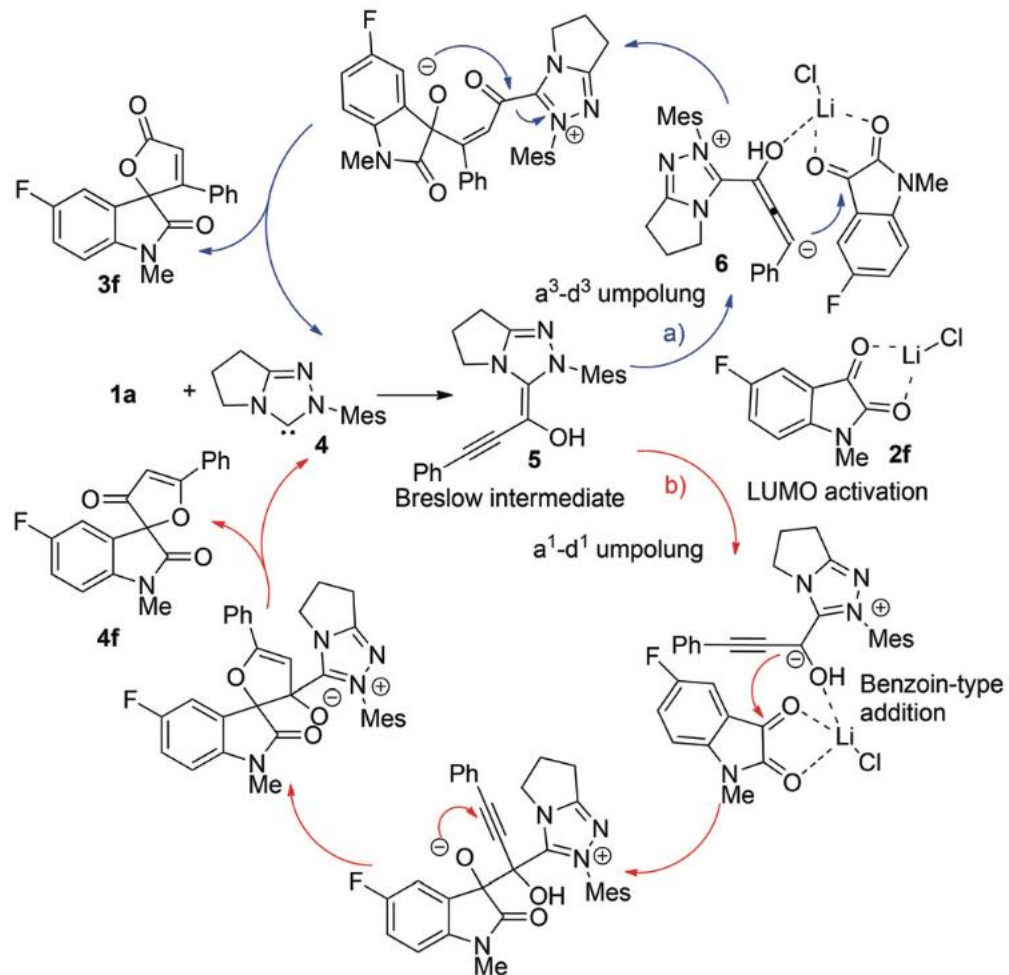
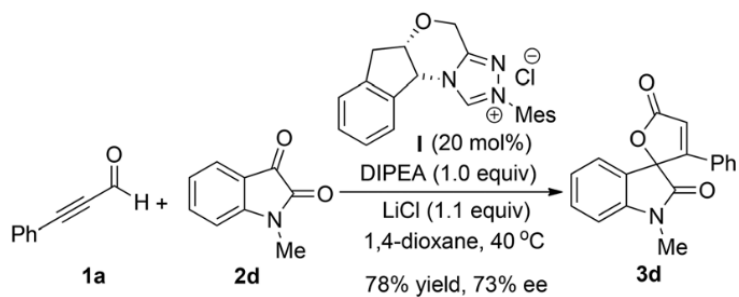


2.1 NHC/Lewis Acid Cooperative Catalysis

Annulation of ynals and isatins :

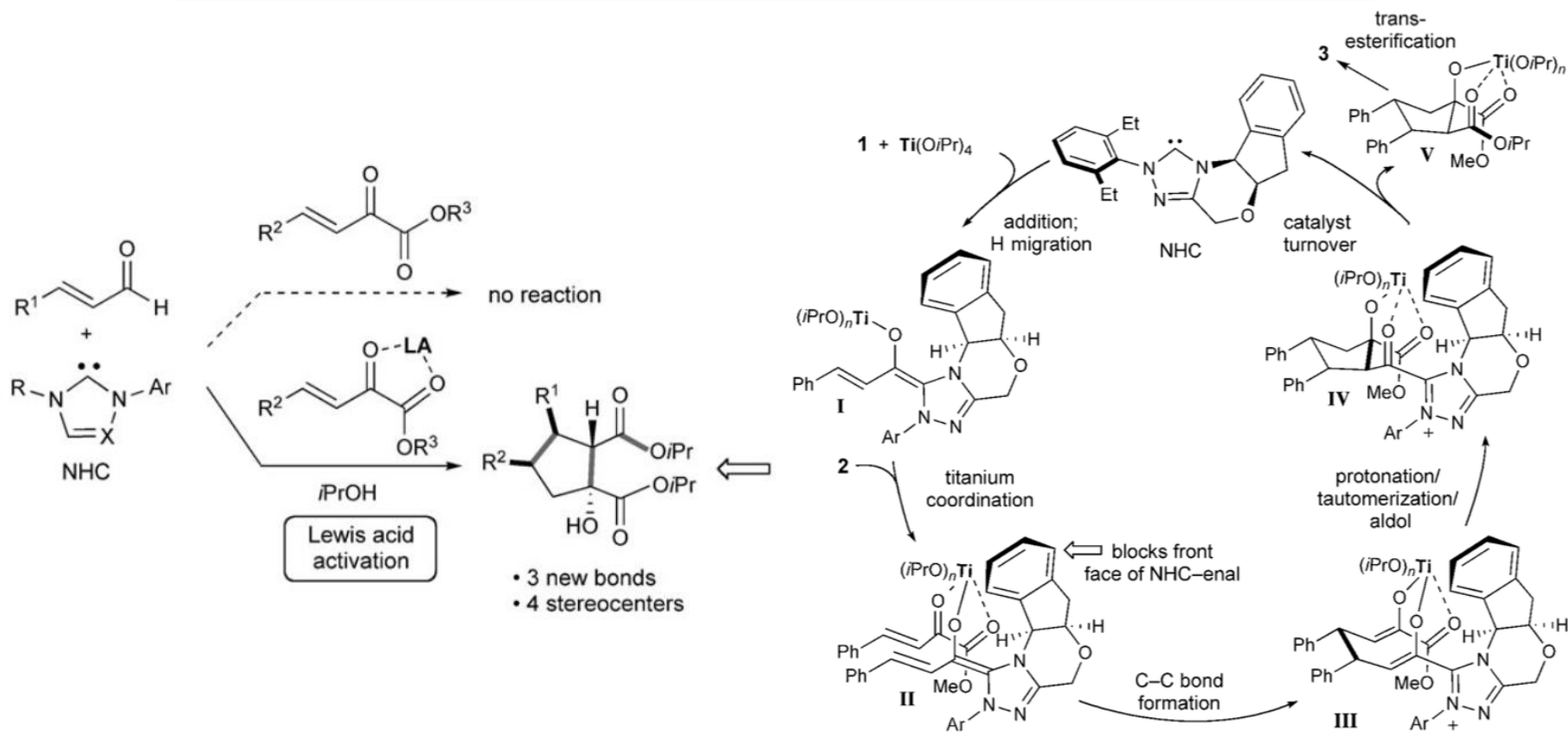
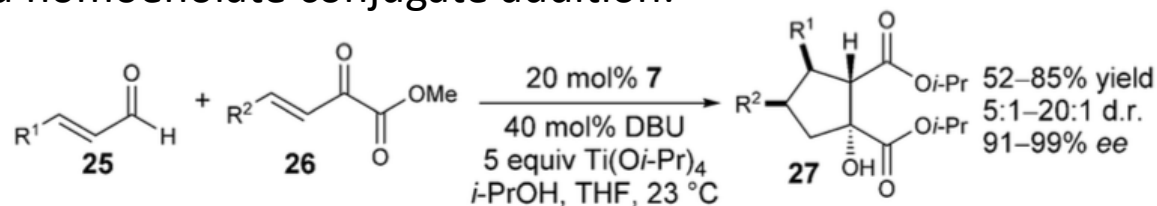


2.1 NHC/Lewis Acid Cooperative Catalysis



2.1 NHC/Lewis Acid Cooperative Catalysis

NHC/Lewis acid homoenolate conjugate addition:



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2.4 NHC/Brønsted Acid Cooperative Catalysis

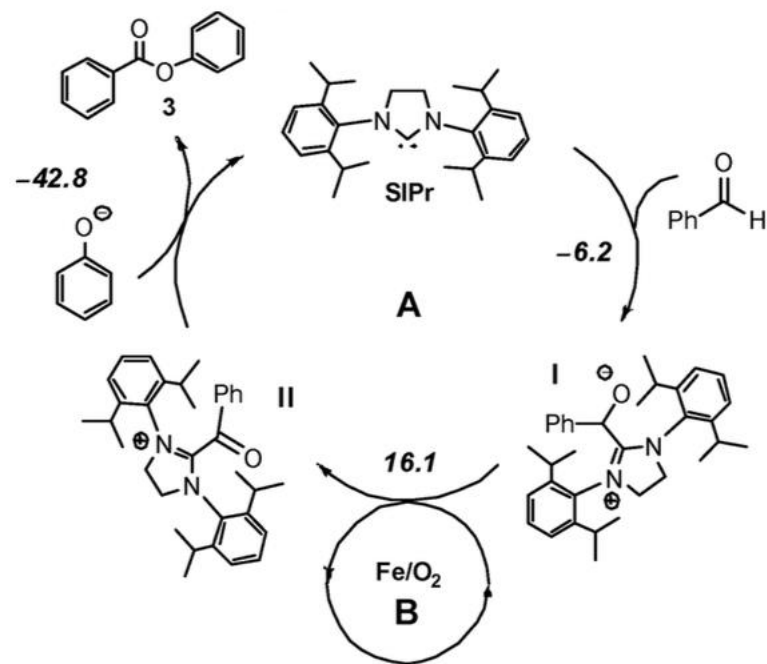
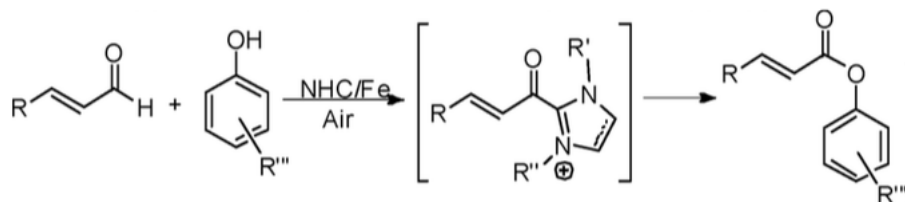
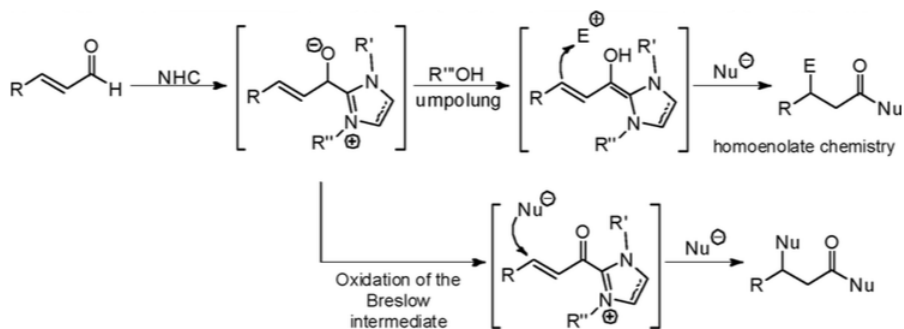
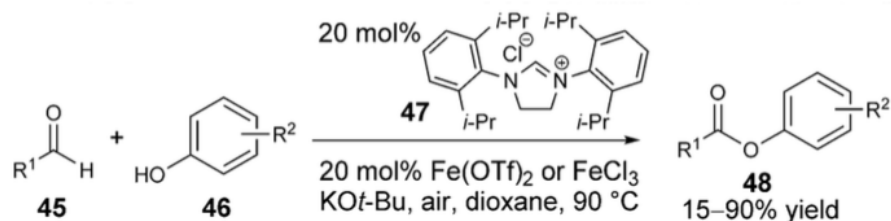
2.5 NHC/Brønsted Base Cooperative Catalysis

3. Summary and Outlook

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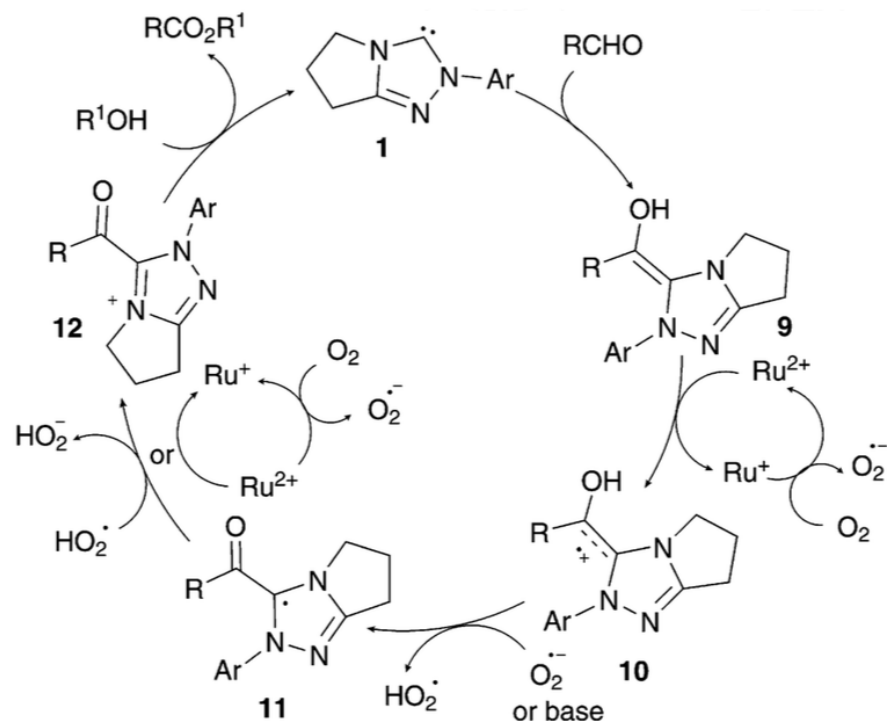
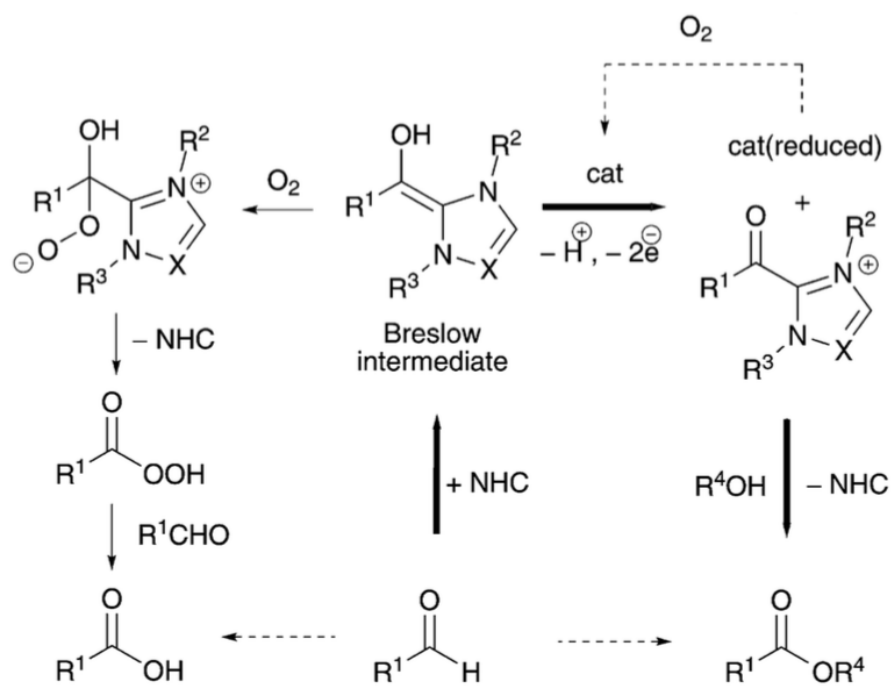
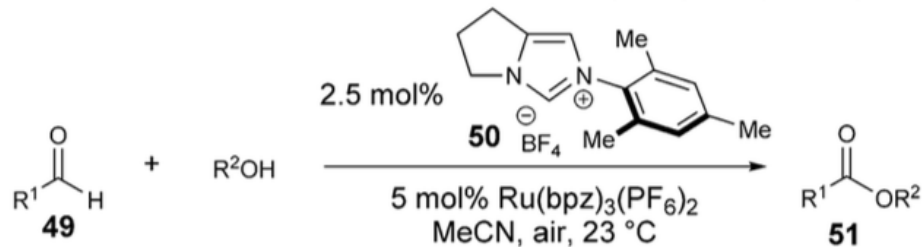
2.2 Oxidative NHC Catalysis

NHC/iron-catalyzed esterification :



2.2 Oxidative NHC Catalysis

Ruthenium-catalyzed oxidation of Breslow intermediates :



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2. NHC Cooperative Catalysis

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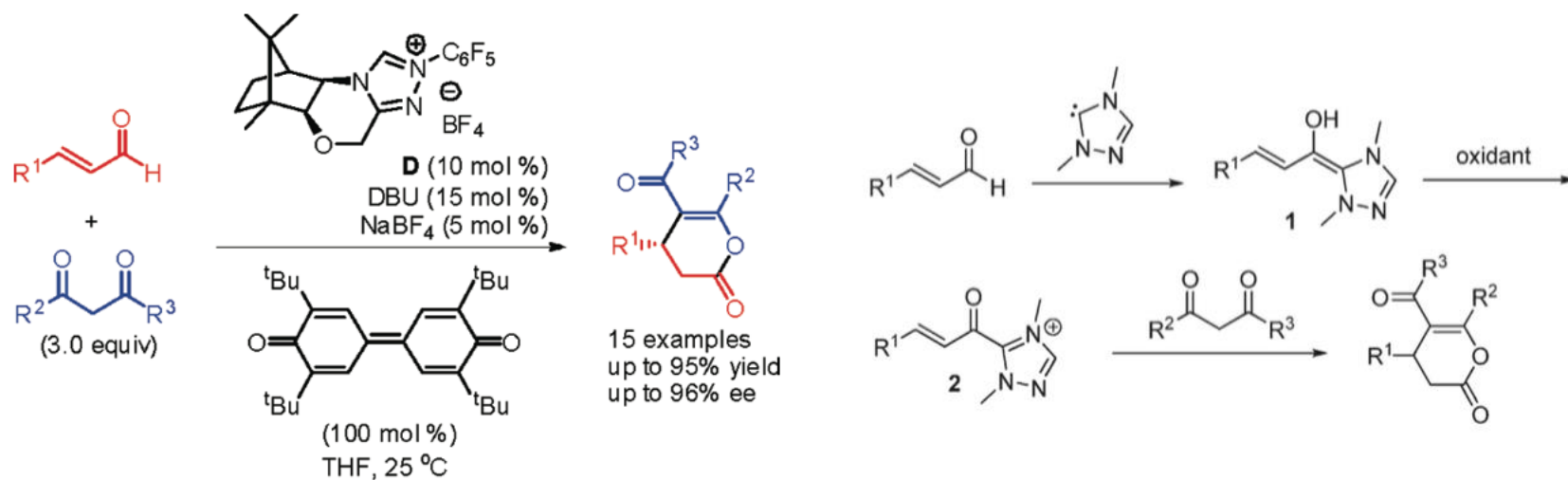
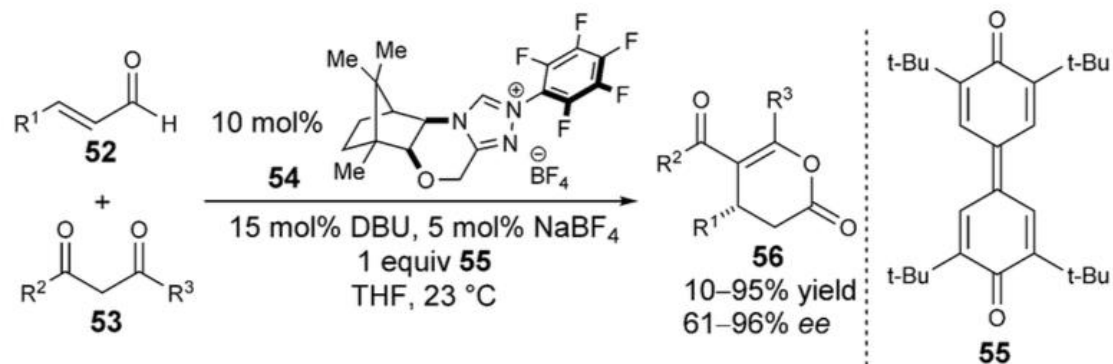
2.5 NHC/Brønsted Base Cooperative Catalysis

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2.3 Oxidative NHC/Lewis Acid Cooperative Catalysis

Oxidative NHC/Lewis acid catalyzed annulation :



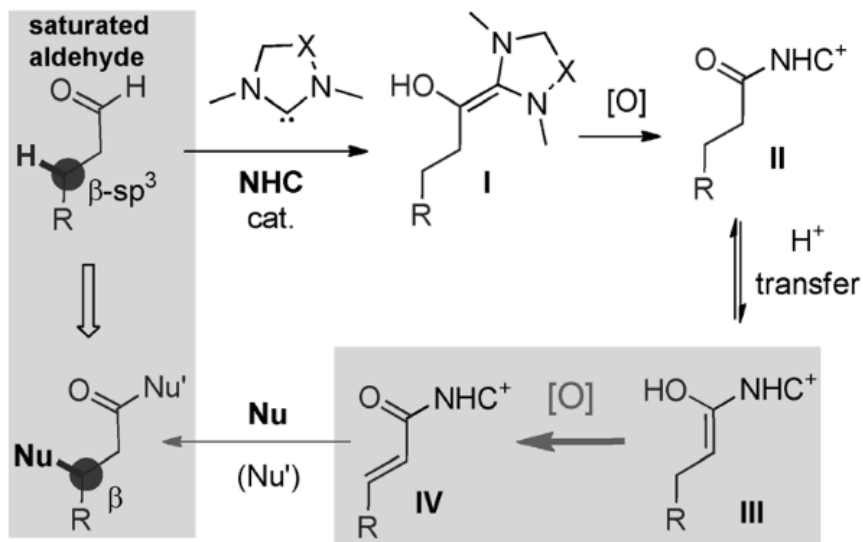
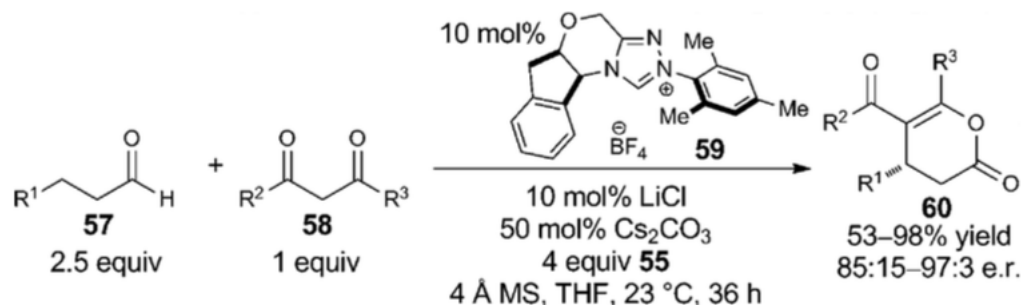
Z.-Q. Rong, M.-Q. Jia, S.-L. You, *Org. Lett.* 2011, 13, 4080

De Sarkar, S.; Studer, A. *Angew. Chem., Int. Ed.* 2010, 49, 9266

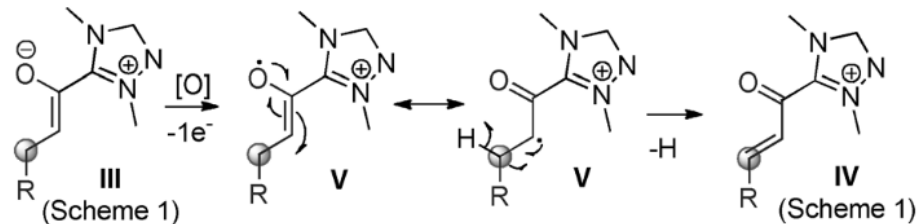
De Sarkar, S.; Studer, A. *Org. Lett.* 2010, 12, 1992

2.3 Oxidative NHC/Lewis Acid Cooperative Catalysis

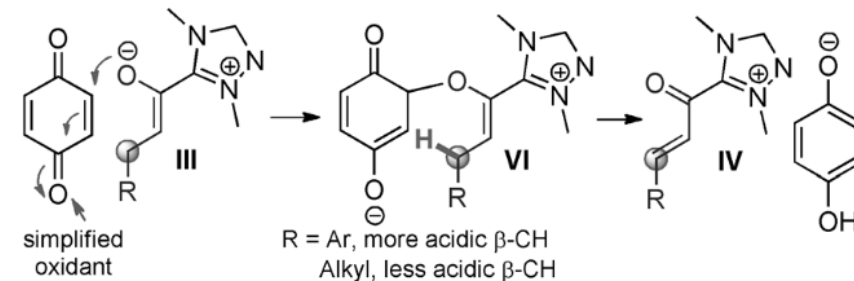
Oxidative NHC/Lewis acid catalyzed annulation with saturated aldehydes :



a) single-electron-transfer (radical) mechanism:

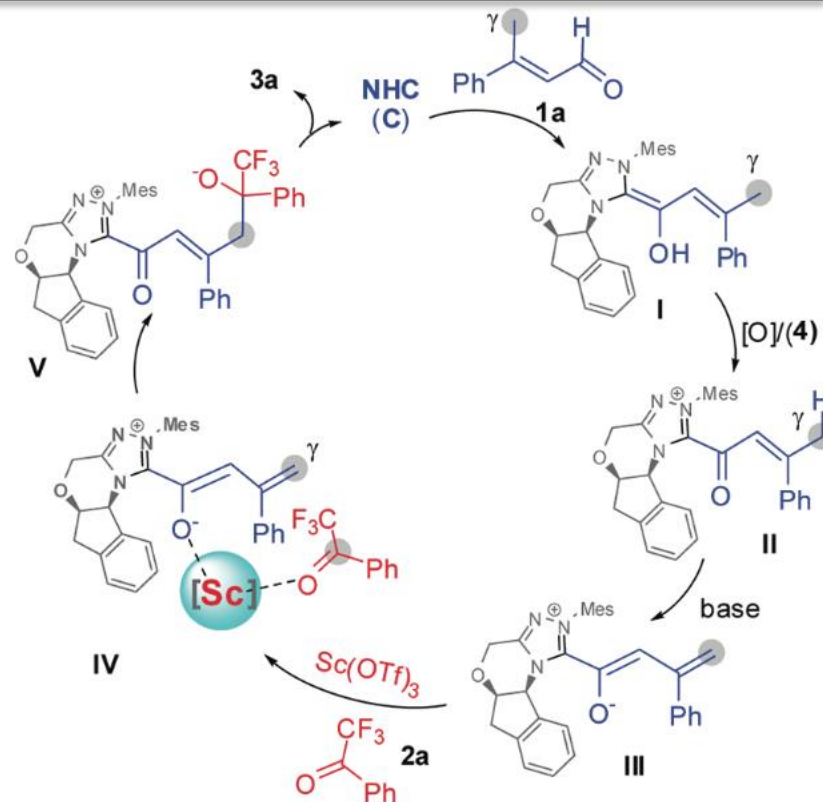
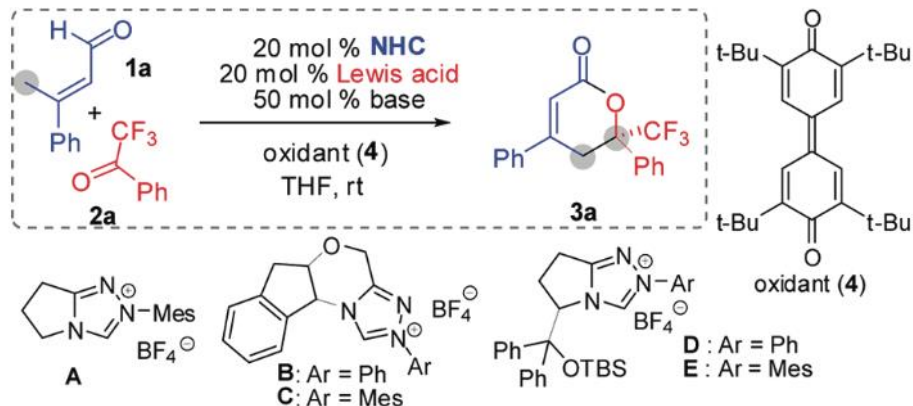
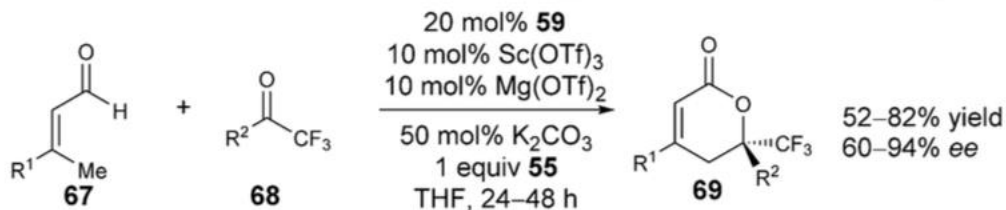


b) electron-pair-transfer mechanism:



2.3 Oxidative NHC/Lewis Acid Cooperative Catalysis

Oxidative NHC/Lewis acid catalyzed γ -addition :



J. Mo, X. Chen, Y. R. Chi, J. Am. Chem. Soc. 2012, 134, 8810

R. Liu, C. Yu, Z. Xiao, T. Li, X. Wang, Y. Xie, C. Yao, Org. Biomol. Chem. 2014, 12, 1885

Outline:

1. Introduction

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2.4 NHC/Brønsted Acid Cooperative Catalysis

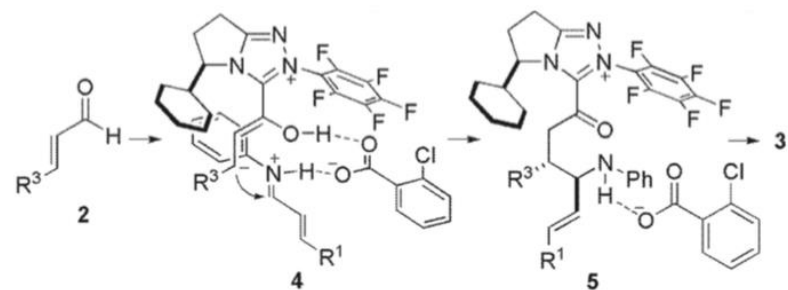
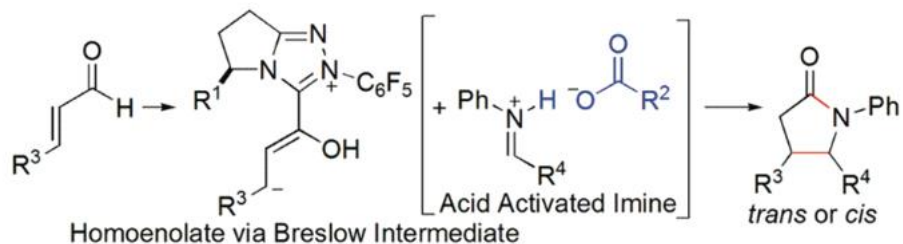
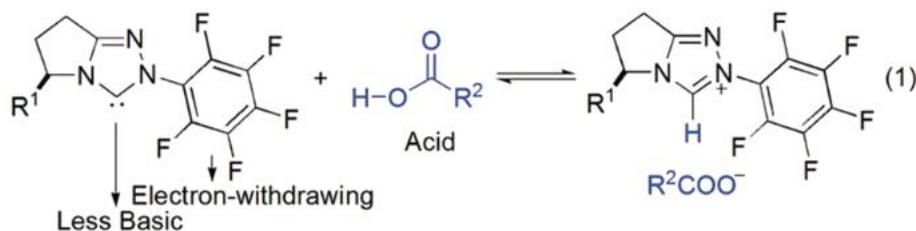
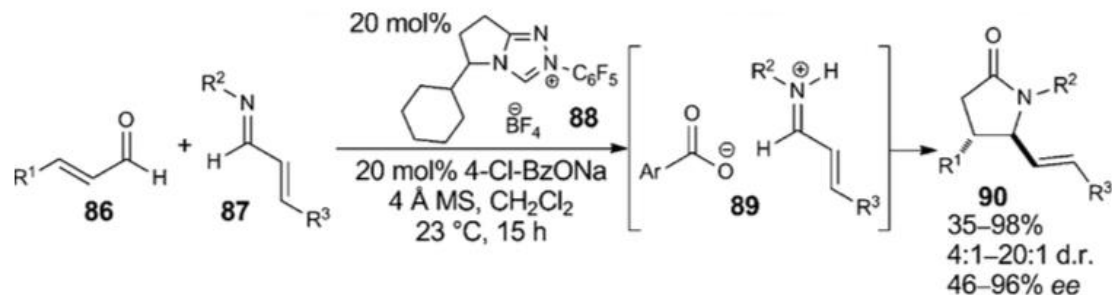
2.5 NHC/Brønsted Base Cooperative Catalysis

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2.4 NHC/Brønsted Acid Cooperative Catalysis

NHC/Brønsted acid cooperative catalysis :



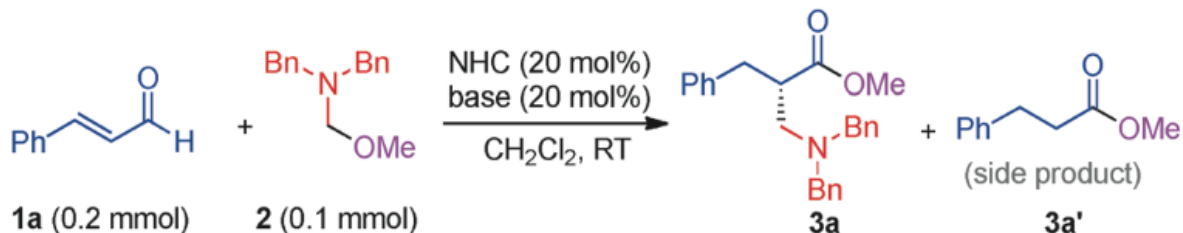
Proposed Pathway for *trans*- γ -Lactam Formation

X. Zhao, D. A. DiRocco, T. Rovis, J. Am. Chem. Soc. 2011, 133, 12466

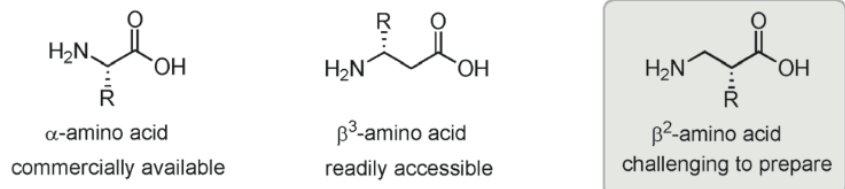
Y. Que, Y. Lu, W. Wang, Y. Wang, H. Wang, C. Yu, T. Li, X.-S. Wang, S. Shen, C. Yao, Chem. Asian J. 2016, 11, 678.

2.4 NHC/Brønsted Acid Cooperative Catalysis

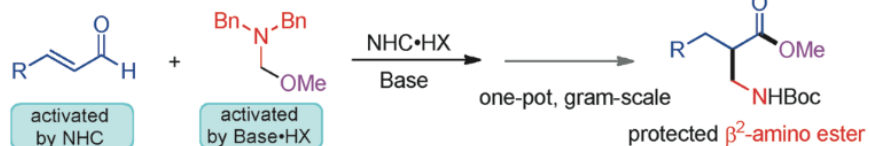
Synthesis of β -amino esters :



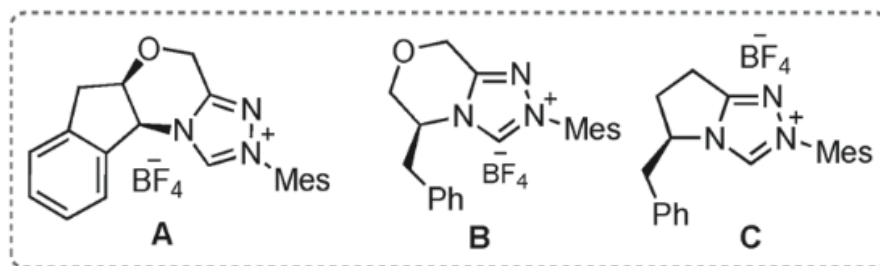
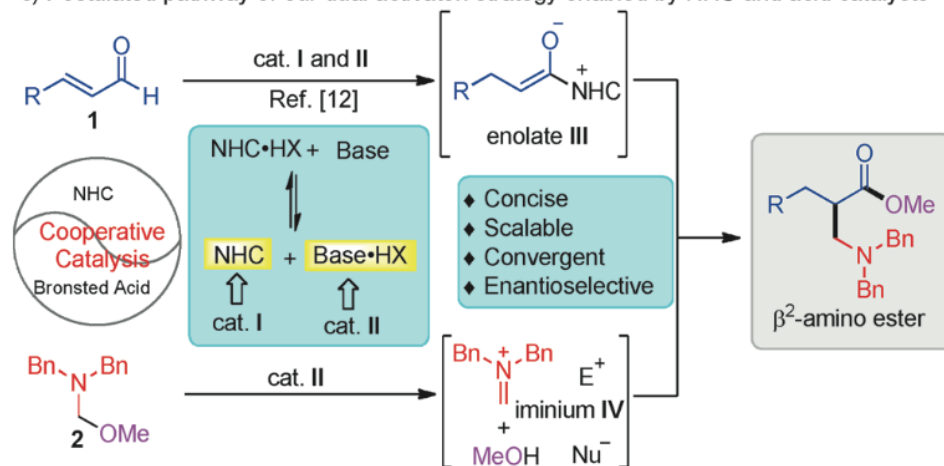
a) Classification of amino acids



b) Our catalytic approach for rapid access to β^2 -amino acids



c) Postulated pathway of our dual activation strategy enabled by NHC and acid catalysts



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3 Summary and Outlook

1. The field of NHC catalysis is still relatively new, and has seen enormous growth in terms of substrate scope, reaction conditions, and new modes of reactivity since 2004.
2. This last vein of research has especially revitalized interest in carbene catalysis and revealed new possible reaction types not previously available to NHC catalysis alone.
3. NHCs have been combined with Lewis acids/bases, oxidants, Brønsted acid/base. NHCs were initially isolated and studied as distinct catalysts owing to key inspiration from biological systems.
4. Continued studies by the worldwide carbene catalysis community will undoubtedly provide new means of reaction discovery and development.

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4 Acknowledgement

Prof. Huang

Meng sixuan

All members in E201

Everyone here

*Thanks for your
attention!*